

# Rocks and Minerals







Goal: Students will be able to describe the characteristics an object must possess to be classified as a mineral.



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# What makes up an environment?

- **Living** - plants, animals, bacteria, etc.
- **Non-living** - rocks, minerals, man-made materials.



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January  
Garnet



February  
Amethyst



March  
Aquamarine



April  
Crystal



May  
Emerald



June  
Alexandrite



July  
Ruby



August  
Peridot



September  
Sapphire



October  
Rose Zircon



November  
Topaz



December  
Blue Zircon



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# Minerals

- Natural (not man-made).
- Non-living (inorganic).
- Solid, with a crystalline atomic structure. (Atoms are arranged.)
- Definite composition. (Certain elements or compounds.)



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Periodic Table of the Elements																		18
1	2											13	14	15	16	17	18	
1A	2A											3A	4A	5A	6A	7A	8A	
11A	2A											3A	4A	5A	6A	7A	8A	
1	2											5	6	7	8	9	10	
H	He											B	C	N	O	F	Ne	
Hydrogen	Helium											Boron	Carbon	Nitrogen	Oxygen	Fluorine	Neon	
1.0079	4.00260											10.811	12.011	14.00674	15.9994	18.998403	20.1797	
3	4											13	14	15	16	17	18	
Li	Be											Al	Si	P	S	Cl	Ar	
Lithium	Beryllium											Aluminum	Silicon	Phosphorus	Sulfur	Chlorine	Argon	
6.941	9.01218											26.981539	28.0855	30.973762	32.066	35.4527	39.948	
11	12	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	
Na	Mg	IIIB	IVB	VB	VIB	VIIB	VIII			IB	IIB	Al	Si	P	S	Cl	Ar	
Sodium	Magnesium	3B	4B	5B	6B	7B	8			1B	2B	Aluminum	Silicon	Phosphorus	Sulfur	Chlorine	Argon	
22.989768	24.305											26.981539	28.0855	30.973762	32.066	35.4527	39.948	
19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36	
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr	
Potassium	Calcium	Scandium	Titanium	Vanadium	Chromium	Manganese	Iron	Cobalt	Nickel	Copper	Zinc	Gallium	Germanium	Arsenic	Selenium	Bromine	Krypton	
39.0983	40.078	44.95591	47.88	50.9415	51.9961	54.938	55.847	58.9332	58.6934	63.546	65.39	69.732	72.64	74.92159	78.96	79.904	83.80	
37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54	
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe	
Rubidium	Strontium	Yttrium	Zirconium	Niobium	Molybdenum	Technetium	Ruthenium	Rhodium	Palladium	Silver	Cadmium	Indium	Tin	Antimony	Tellurium	Iodine	Xenon	
85.4678	87.62	88.90585	91.224	92.90638	95.94	98.9072	101.07	102.9055	106.42	107.8682	112.411	114.818	118.71	121.760	127.6	126.90447	131.29	
55	56	57-71	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86	
Cs	Ba		Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn	
Cesium	Barium		Hafnium	Tantalum	Tungsten	Rhenium	Osmium	Iridium	Platinum	Gold	Mercury	Thallium	Lead	Bismuth	Polonium	Astatine	Radon	
132.90543	137.327		178.49	180.9479	183.85	186.207	190.23	192.22	195.08	196.9665	200.59	204.3833	207.2	208.98037	[208.9824]	209.9871	222.0176	
87	88	89-103	104	105	106	107	108	109	110	111	112	113	114	115	116	117	118	
Fr	Ra		Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	Cn	Uut	Uuq	Uup	Uuh	Uus	Uuo	
Francium	Radium		Rutherfordium	Dubnium	Seaborgium	Bohrium	Hassium	Meitnerium	Darmstadtium	Roentgenium	Copernicium	Ununtrium	Ununquadium	Ununpentium	Ununhexium	Ununseptium	Ununoctium	
223.0197	226.0254		[261]	[262]	[266]	[264]	[269]	[268]	[269]	[272]	[277]	unknown	[289]	unknown	[298]	unknown	unknown	
Lanthanide Series		57	58	59	60	61	62	63	64	65	66	67	68	69	70	71		
		La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu		
		Lanthanum	Cerium	Praseodymium	Neodymium	Promethium	Samarium	Europium	Gadolinium	Terbium	Dysprosium	Holmium	Erbium	Thulium	Ytterbium	Lutetium		
		138.9055	140.115	140.90765	144.24	144.9127	150.36	151.9655	157.25	158.92534	162.50	164.93032	167.26	168.93421	173.04	174.967		
Actinide Series		89	90	91	92	93	94	95	96	97	98	99	100	101	102	103		
		Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr		
		Actinium	Thorium	Protactinium	Uranium	Neptunium	Plutonium	Americium	Curium	Berkelium	Californium	Einsteinium	Fermium	Mendelevium	Nobelium	Lawrencium		
		227.0278	232.0381	231.03588	238.0289	237.0482	244.0642	243.0614	247.0703	247.0703	251.0796	[254]	257.0951	258.1	259.1009	[262]		
Alkali Metal		Alkaline Earth		Transition Metal		Basic Metal		Semimetals		Nonmetals		Halogens		Noble Gas		Lanthanides		Actinides



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Goal: Students will be able to use different properties of minerals.

# Properties of Minerals



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# Color





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# Streak

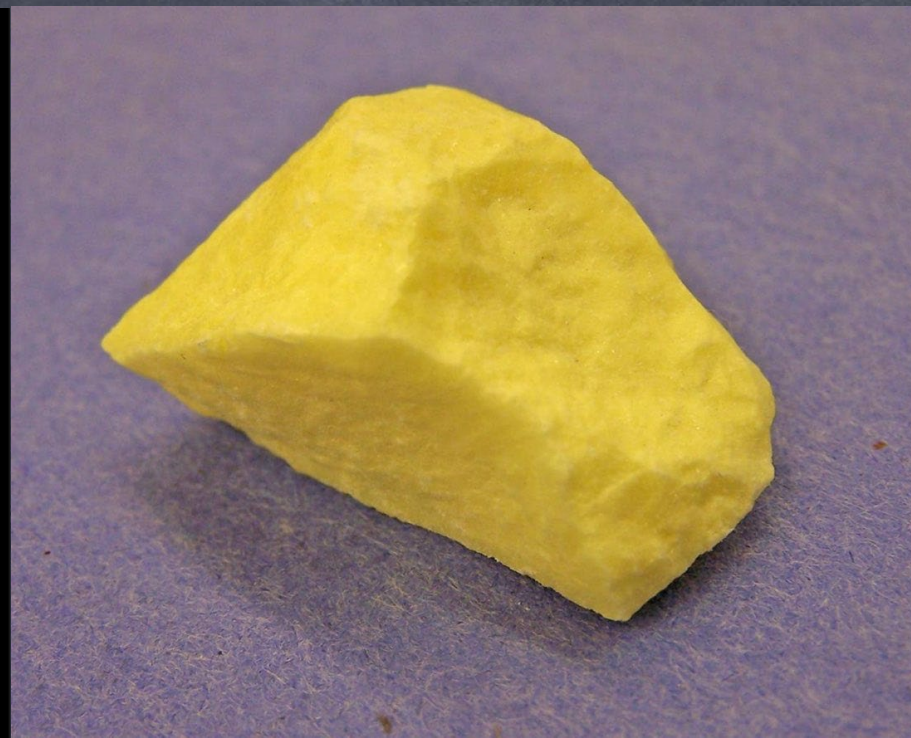




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# Luster

- **Metallic** - reflect light only from their outer surface. Looks like a metal.
- **Non-metallic** - can be glassy, dull, pearly, waxy, etc.

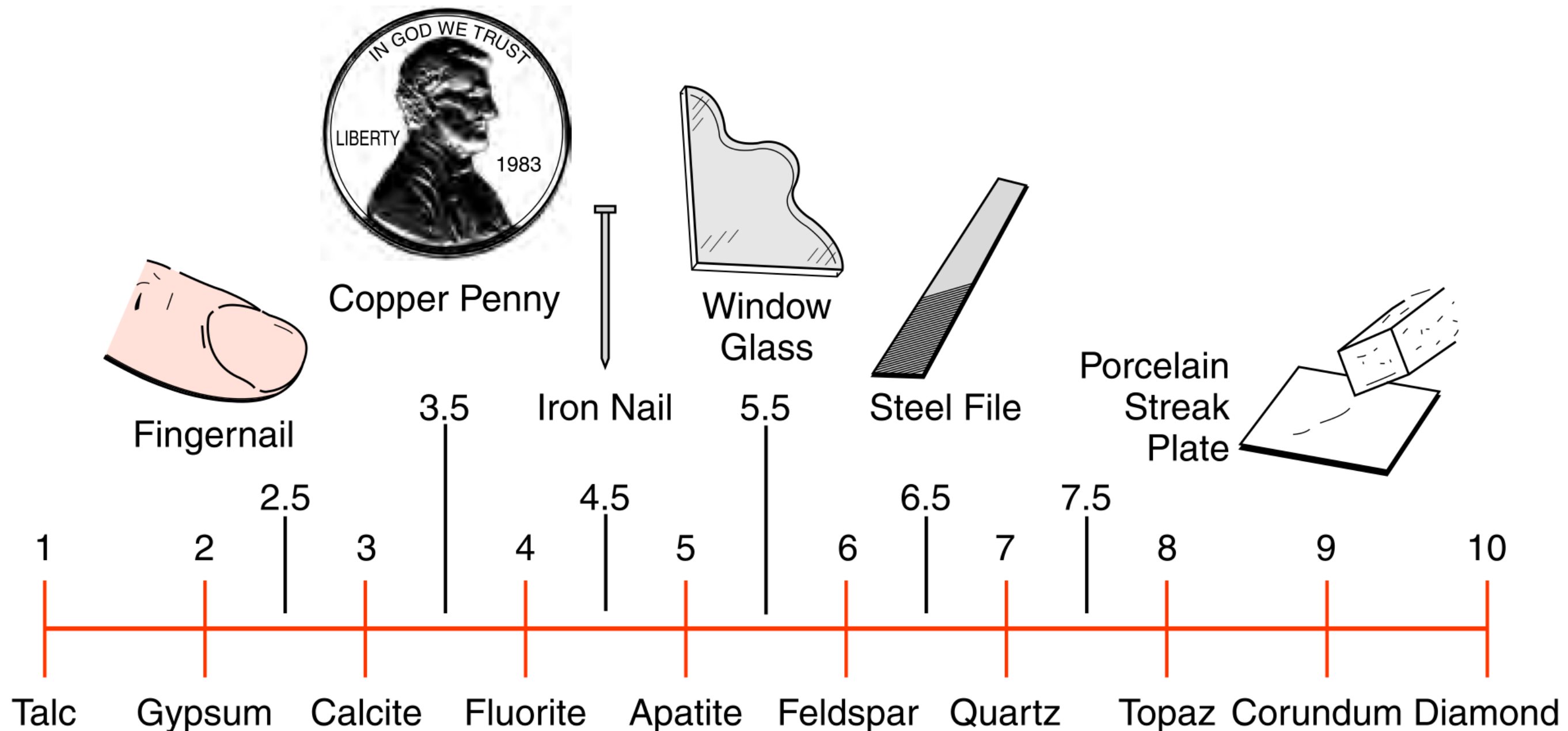




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**Hardness** - resistance to being scratched.

**Moh's Hardness Scale** - ranges from 1 (**softest**) to 10 (**hardest**).

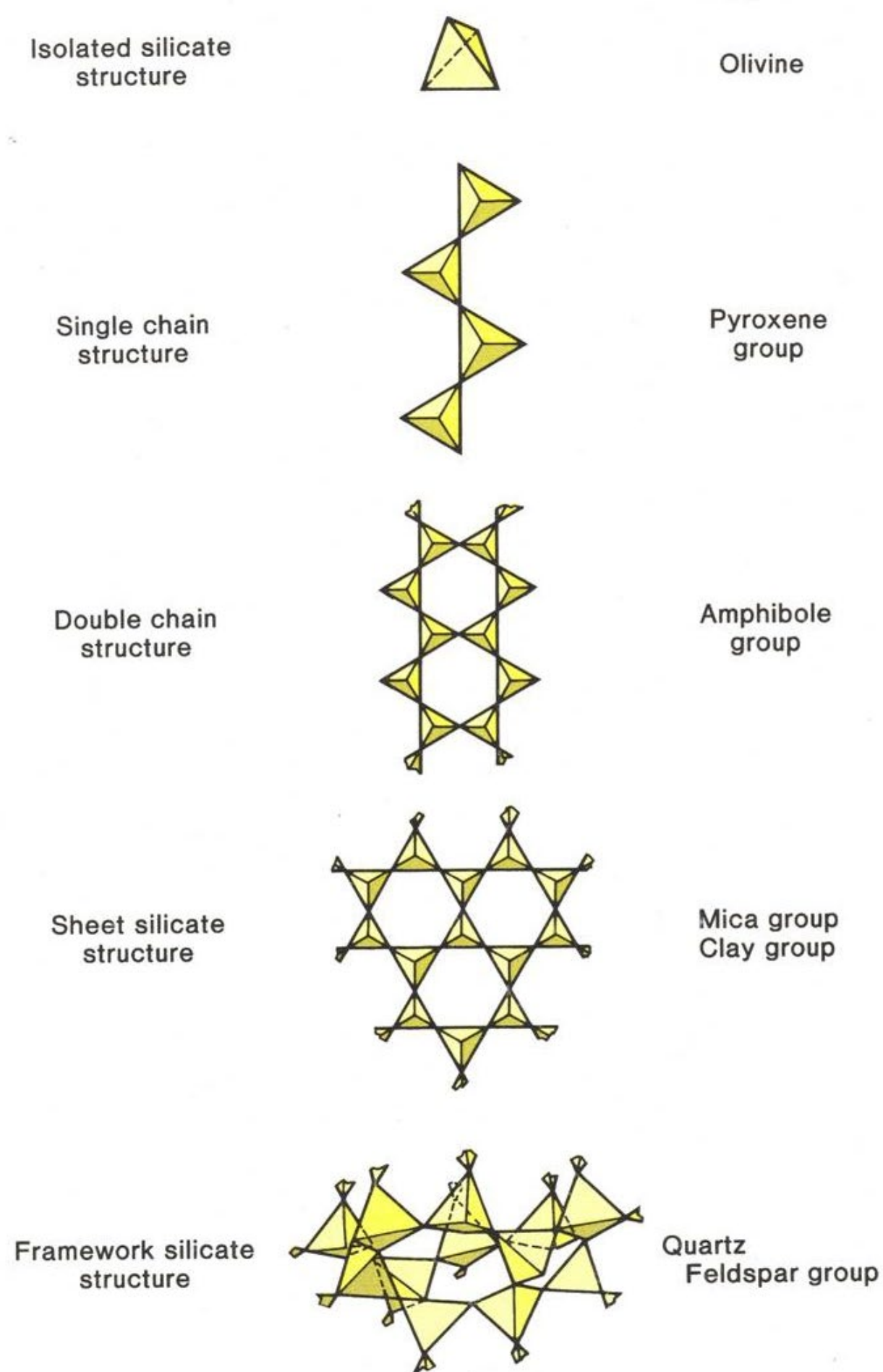




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# Crystal Shape

- The way atoms arrange controls the shape and properties of the mineral.
- Crystal shape is very specific to a mineral.
- Substances that share the same composition can still be different minerals because of their crystal structure.

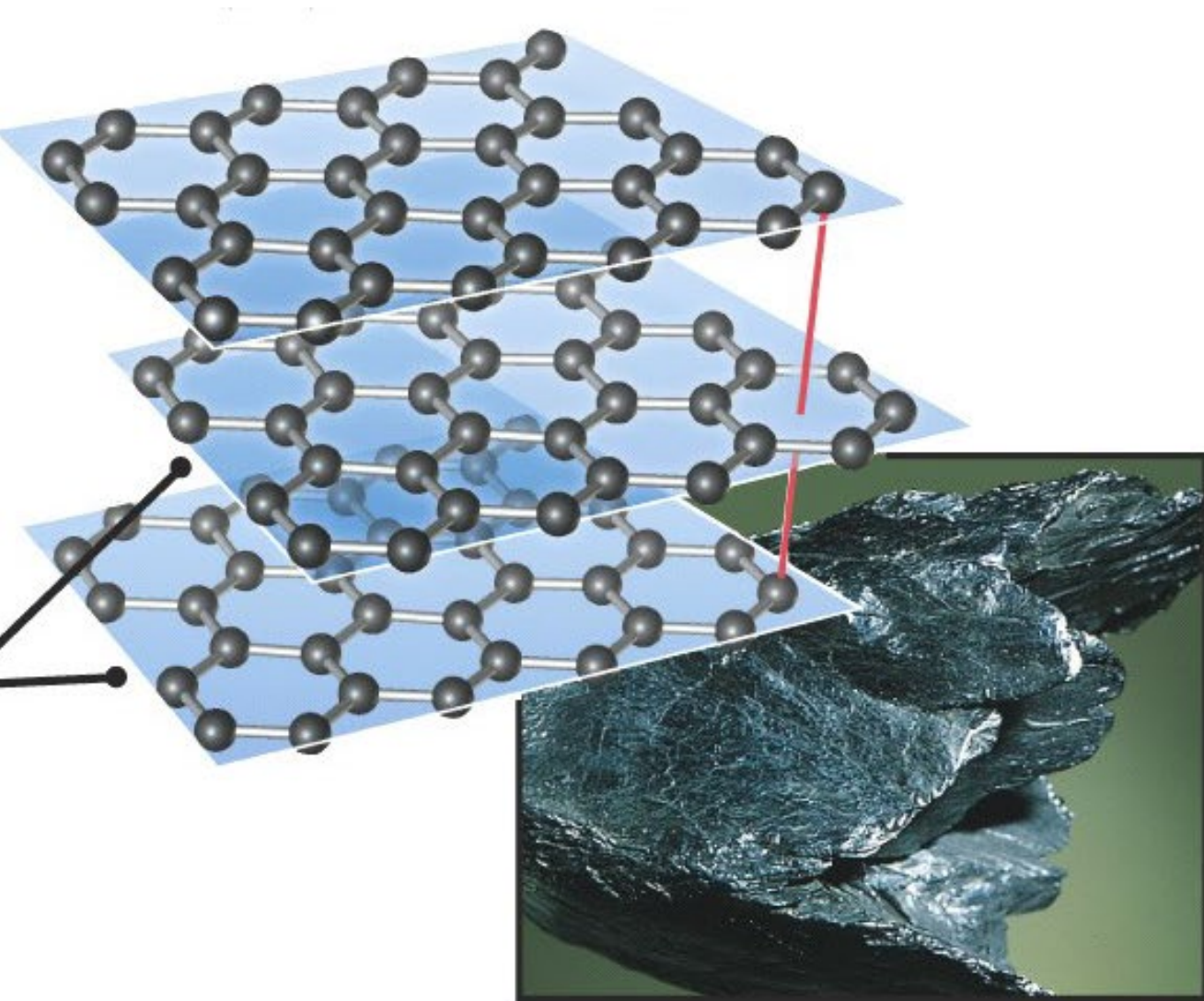




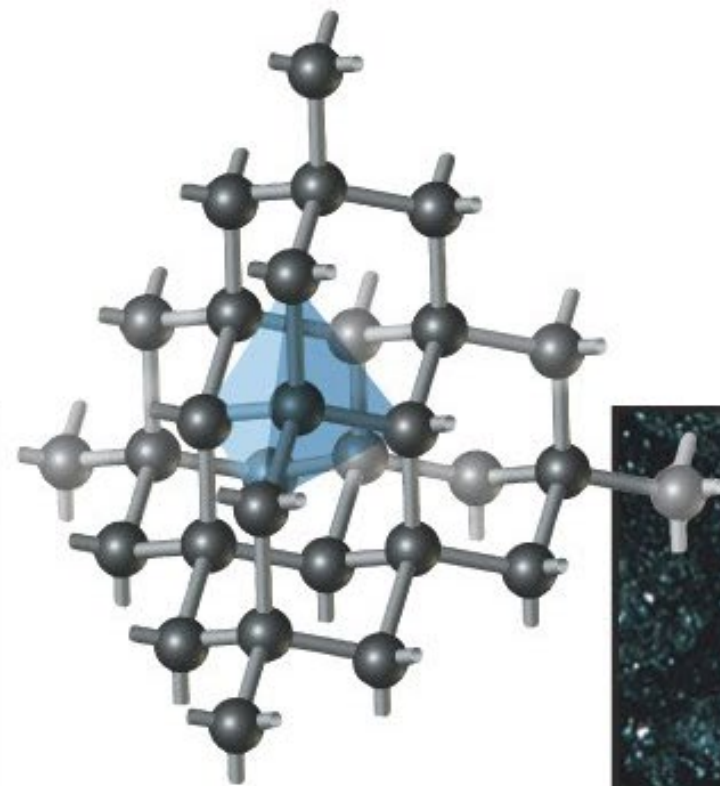




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**Graphite**

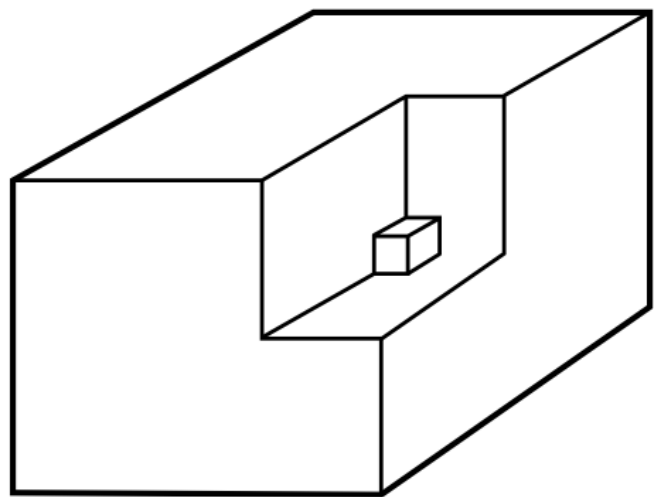


**Diamond**

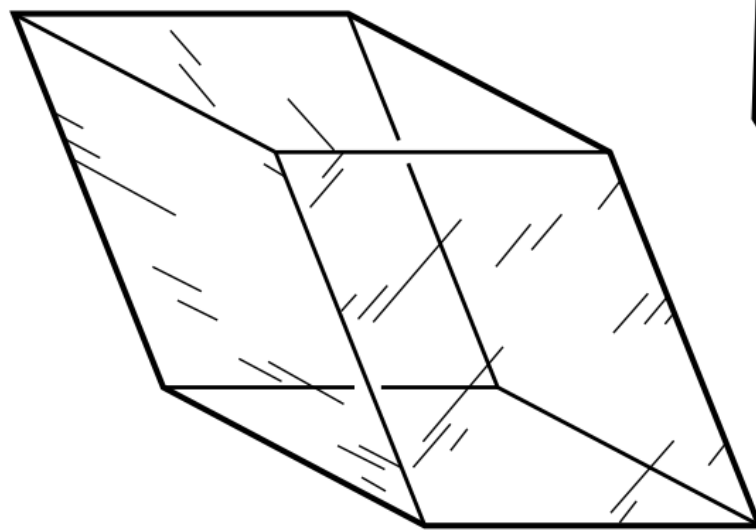


# Cleavage and Fracture

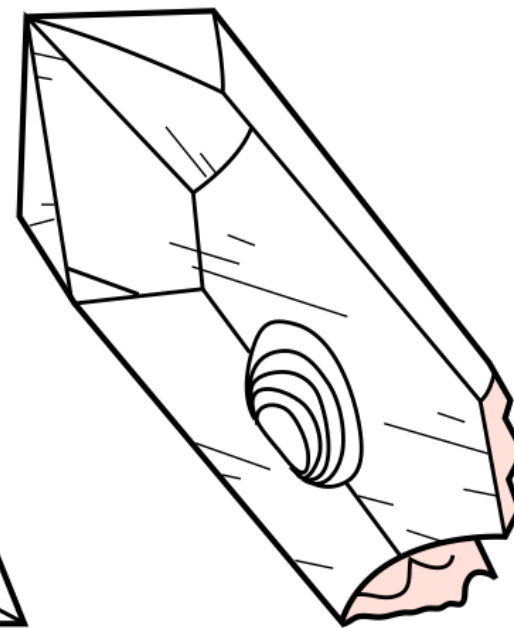
- **Cleavage** - the tendency of some minerals to break along smooth, flat planes.
- **Fracture** - breakage along curved surfaces or surfaces that are not parallel.



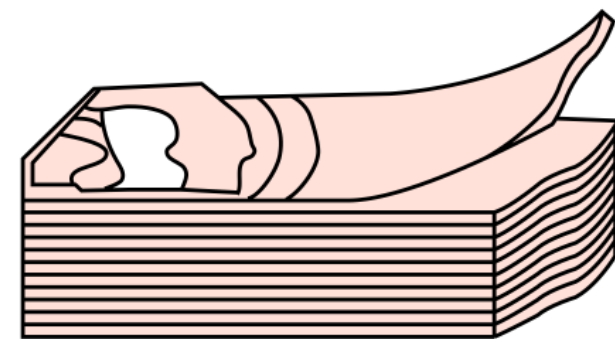
Halite forms cubic crystals that cleave parallel to the crystal faces.



Calcite crystals are rhombohedral and cleave parallel to the crystal faces.



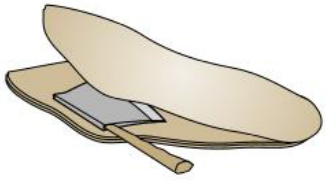
Quartz crystals are hexagonal but fracture along curved surfaces.



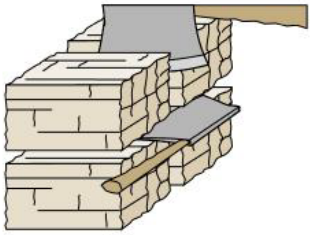
Minerals in the mica family split into thin, flexible sheets.



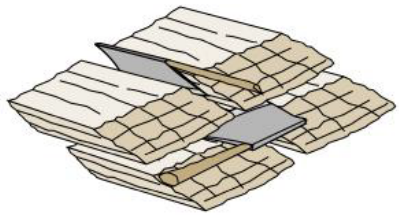
# Goal: Students will be able to use different properties of minerals.



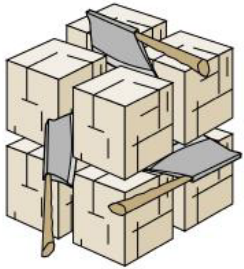
Biotite and muscovite micas exhibit one excellent direction of cleavage.



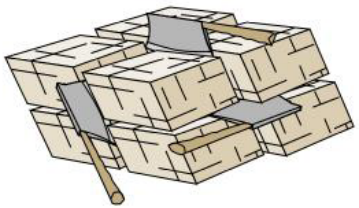
Orthoclase and plagioclase feldspars exhibit two good directions of cleavage at approximately 90deg from each other



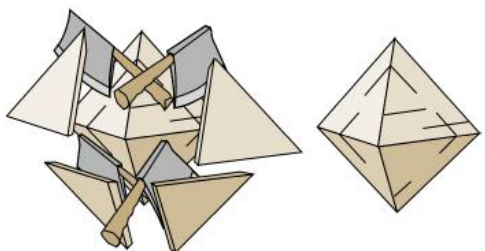
Hornblende (an amphibole) has two good directions of cleavage at 124 deg. from each other.



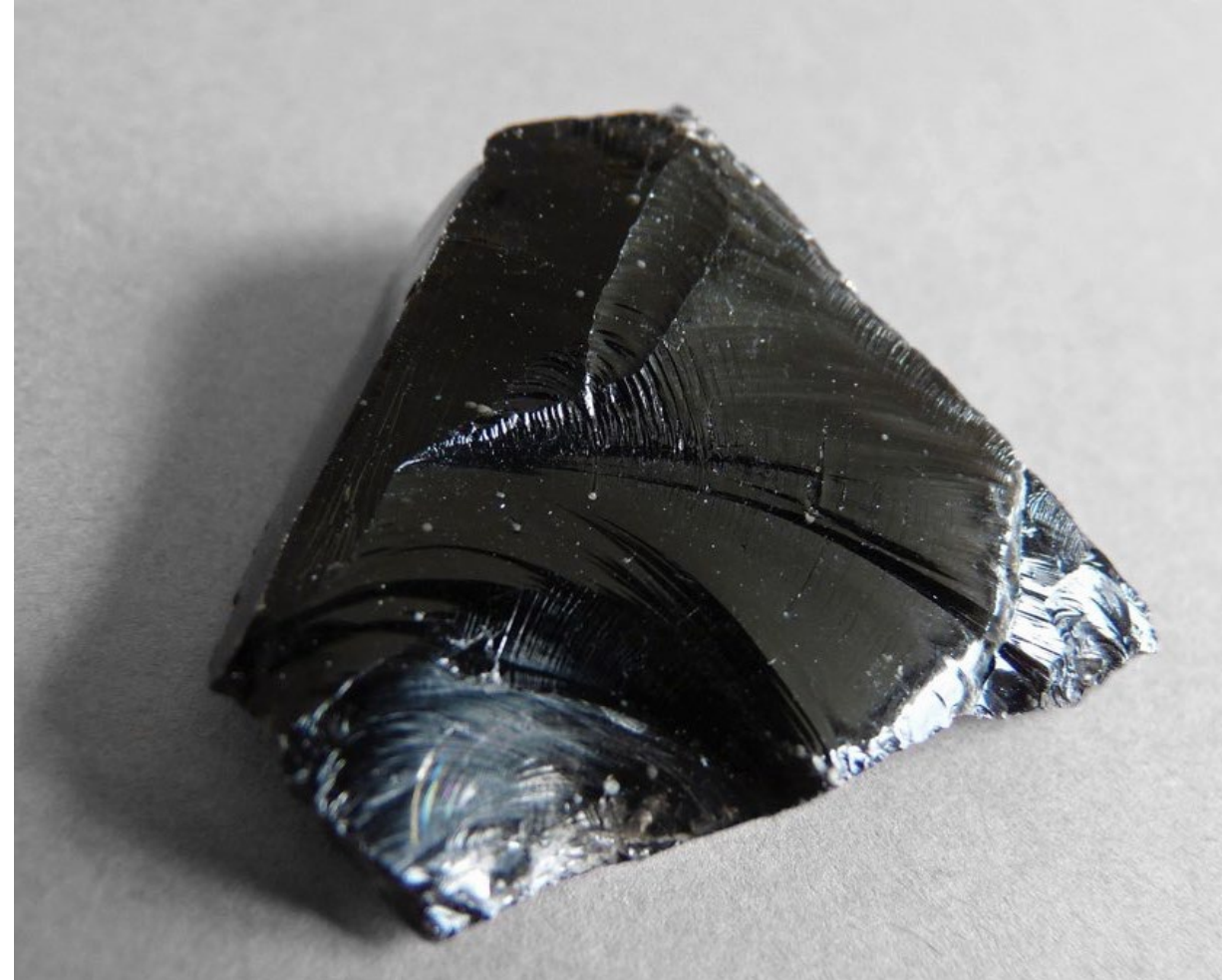
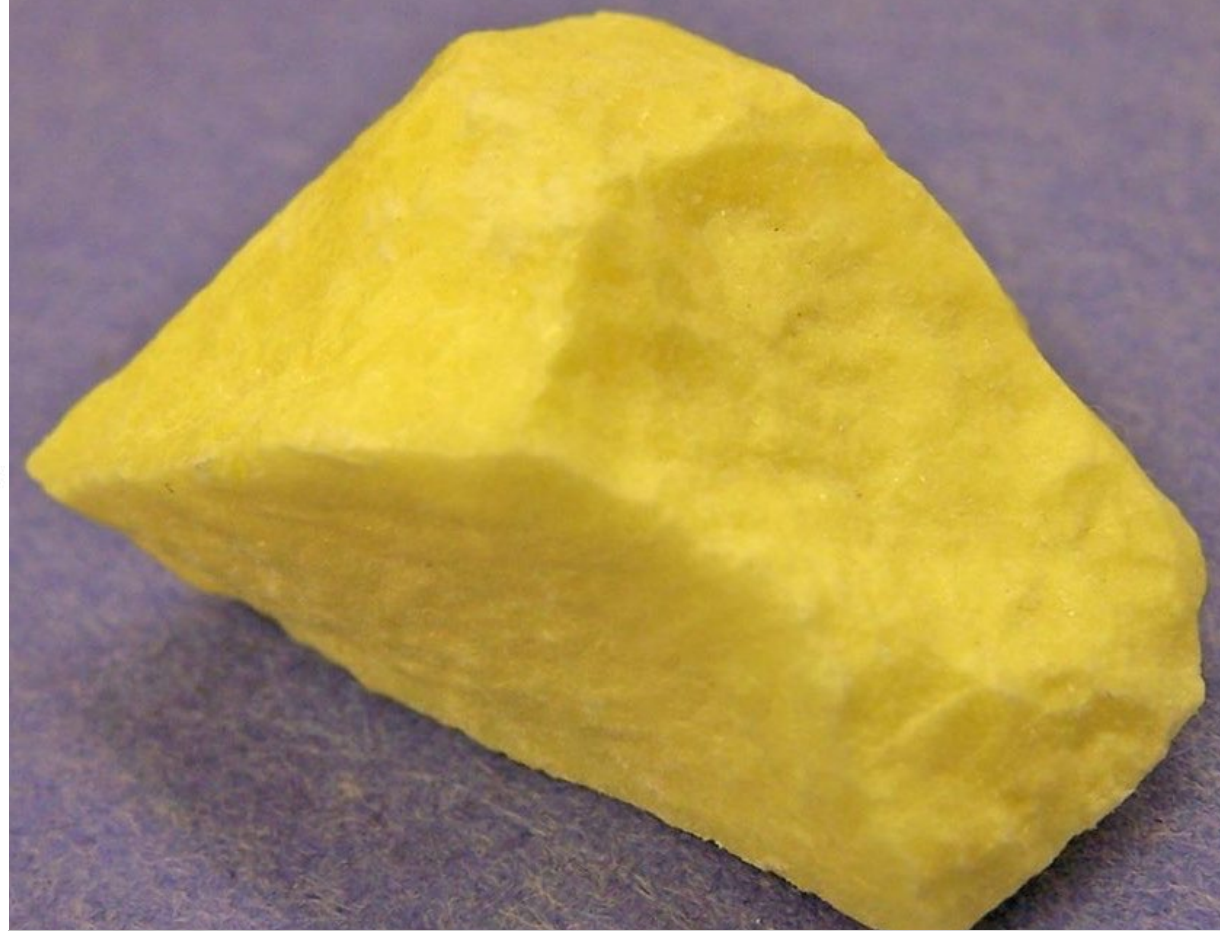
Galena has three good directions of cleavage that form two 90deg. angles (cubic cleavage)



Calcite has three good directions of cleavage that form angles of 105 deg. in one plane and 75 deg. in another



Fluorite has four good directions of cleavage (octahedral cleavage)

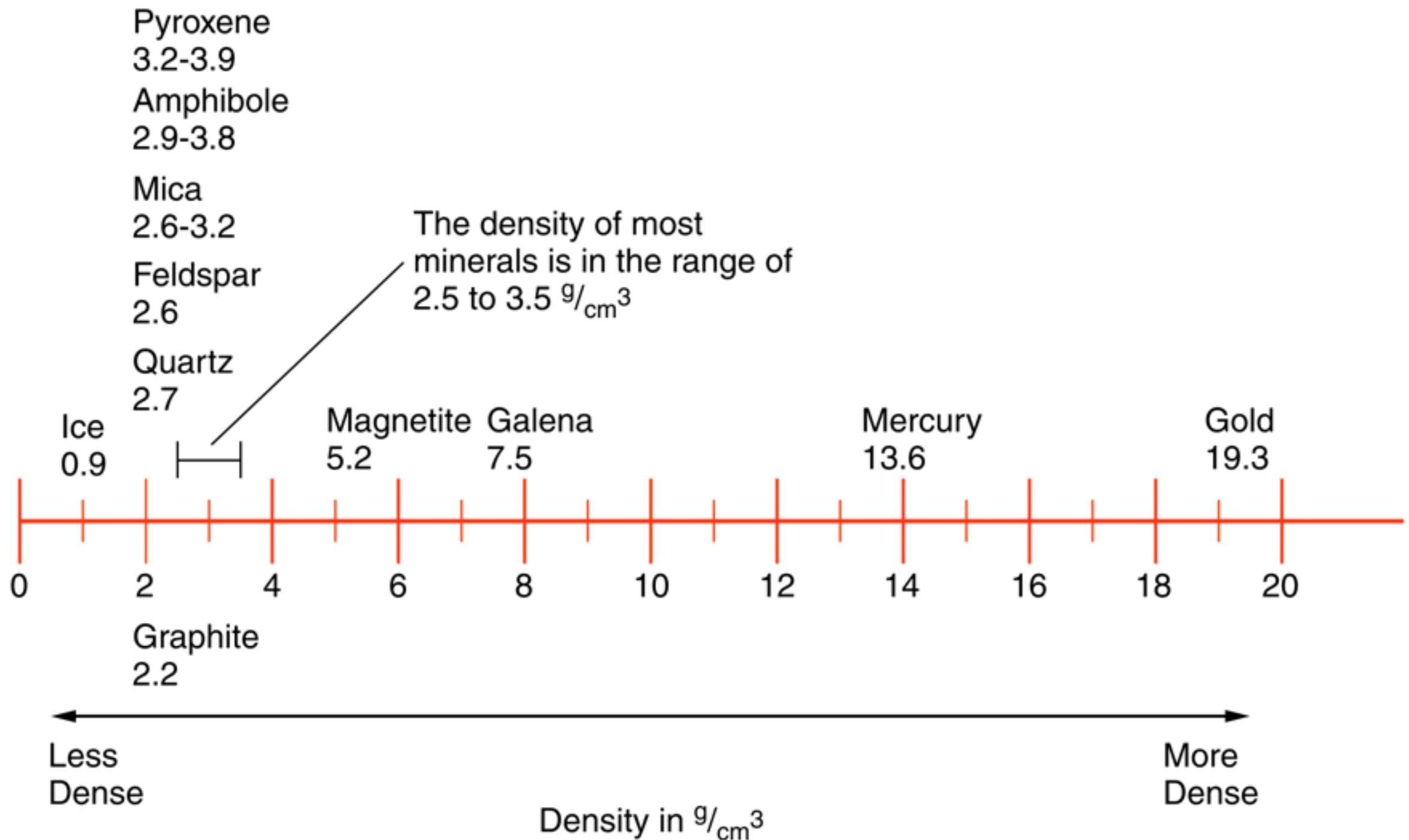




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# Density

$$\text{Density} = \text{Mass} / \text{Volume}$$





# Other Special Properties

- Magnetism
- Taste
- Feel (greasy or slippery)
- Reacts with acids



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Properties of Common Minerals								
LUSTER	HARD- NESS	CLEAVAGE FRACTURE	COMMON COLORS	DISTINGUISHING CHARACTERISTICS	USE(S)	COMPOSITION*	MINERAL NAME	
Metallic luster	1–2	✓	silver to gray	black streak, greasy feel	pencil lead, lubricants	C	Graphite	
	2.5	✓	metallic silver	gray-black streak, cubic cleavage, density = 7.6 g/cm <sup>3</sup>	ore of lead, batteries	PbS	Galena	
	5.5–6.5	✓	black to silver	black streak, magnetic	ore of iron, steel	Fe <sub>3</sub> O <sub>4</sub>	Magnetite	
	6.5	✓	brassy yellow	green-black streak, (fool's gold)	ore of sulfur	FeS <sub>2</sub>	Pyrite	
Either	5.5–6.5 or 1	✓	metallic silver or earthy red	red-brown streak	ore of iron, jewelry	Fe <sub>2</sub> O <sub>3</sub>	Hematite	
Nonmetallic luster	1	✓	white to green	greasy feel	ceramics, paper	Mg <sub>3</sub> Si <sub>4</sub> O <sub>10</sub> (OH) <sub>2</sub>	Talc	
	2	✓	yellow to amber	white-yellow streak	sulfuric acid	S	Sulfur	
	2	✓	white to pink or gray	easily scratched by fingernail	plaster of paris, drywall	CaSO <sub>4</sub> •2H <sub>2</sub> O	Selenite gypsum	
	2–2.5	✓	colorless to yellow	flexible in thin sheets	paint, roofing	KAl <sub>3</sub> Si <sub>3</sub> O <sub>10</sub> (OH) <sub>2</sub>	Muscovite mica	
	2.5	✓	colorless to white	cubic cleavage, salty taste	food additive, melts ice	NaCl	Halite	
	2.5–3	✓	black to dark brown	flexible in thin sheets	construction materials	K(Mg,Fe) <sub>3</sub> AlSi <sub>3</sub> O <sub>10</sub> (OH) <sub>2</sub>	Biotite mica	
	3	✓	colorless or variable	bubbles with acid, rhombohedral cleavage	cement, lime	CaCO <sub>3</sub>	Calcite	
	3.5	✓	colorless or variable	bubbles with acid when powdered	building stones	CaMg(CO <sub>3</sub> ) <sub>2</sub>	Dolomite	
	4	✓	colorless or variable	cleaves in 4 directions	hydrofluoric acid	CaF <sub>2</sub>	Fluorite	
	5–6	✓	black to dark green	cleaves in 2 directions at 90°	mineral collections, jewelry	(Ca,Na)(Mg,Fe,Al)(Si,Al) <sub>2</sub> O <sub>6</sub>	Pyroxene (commonly augite)	
	5.5	✓	black to dark green	cleaves at 56° and 124°	mineral collections, jewelry	CaNa(Mg,Fe) <sub>4</sub> (Al,Fe,Ti) <sub>3</sub> Si <sub>8</sub> O <sub>22</sub> (O,OH) <sub>2</sub>	Amphibole (commonly hornblende)	
	6	✓	white to pink	cleaves in 2 directions at 90°	ceramics, glass	KAlSi <sub>3</sub> O <sub>8</sub>	Potassium feldspar (commonly orthoclase)	
	6	✓	white to gray	cleaves in 2 directions, striations visible	ceramics, glass	(Na,Ca)AlSi <sub>3</sub> O <sub>8</sub>	Plagioclase feldspar	
	6.5	✓	green to gray or brown	commonly light green and granular	furnace bricks, jewelry	(Fe,Mg) <sub>2</sub> SiO <sub>4</sub>	Olivine	
	7	✓	colorless or variable	glassy luster, may form hexagonal crystals	glass, jewelry, electronics	SiO <sub>2</sub>	Quartz	
	6.5–7.5	✓	dark red to green	often seen as red glassy grains in NYS metamorphic rocks	jewelry (NYS gem), abrasives	Fe <sub>3</sub> Al <sub>2</sub> Si <sub>3</sub> O <sub>12</sub>	Garnet	

\*Chemical symbols: Al = aluminum Cl = chlorine H = hydrogen Na = sodium S = sulfur  
 C = carbon F = fluorine K = potassium O = oxygen Si = silicon  
 Ca = calcium Fe = iron Mg = magnesium Pb = lead Ti = titanium

✓ = dominant form of breakage



# Mineral Groups

- Minerals are broken up into groups based on their composition (what they are made of).



Goal: Students will be able to use different properties of minerals.

# What are minerals at Earth's surface made of?

**Average Chemical Composition  
of Earth's Crust, Hydrosphere, and Troposphere**

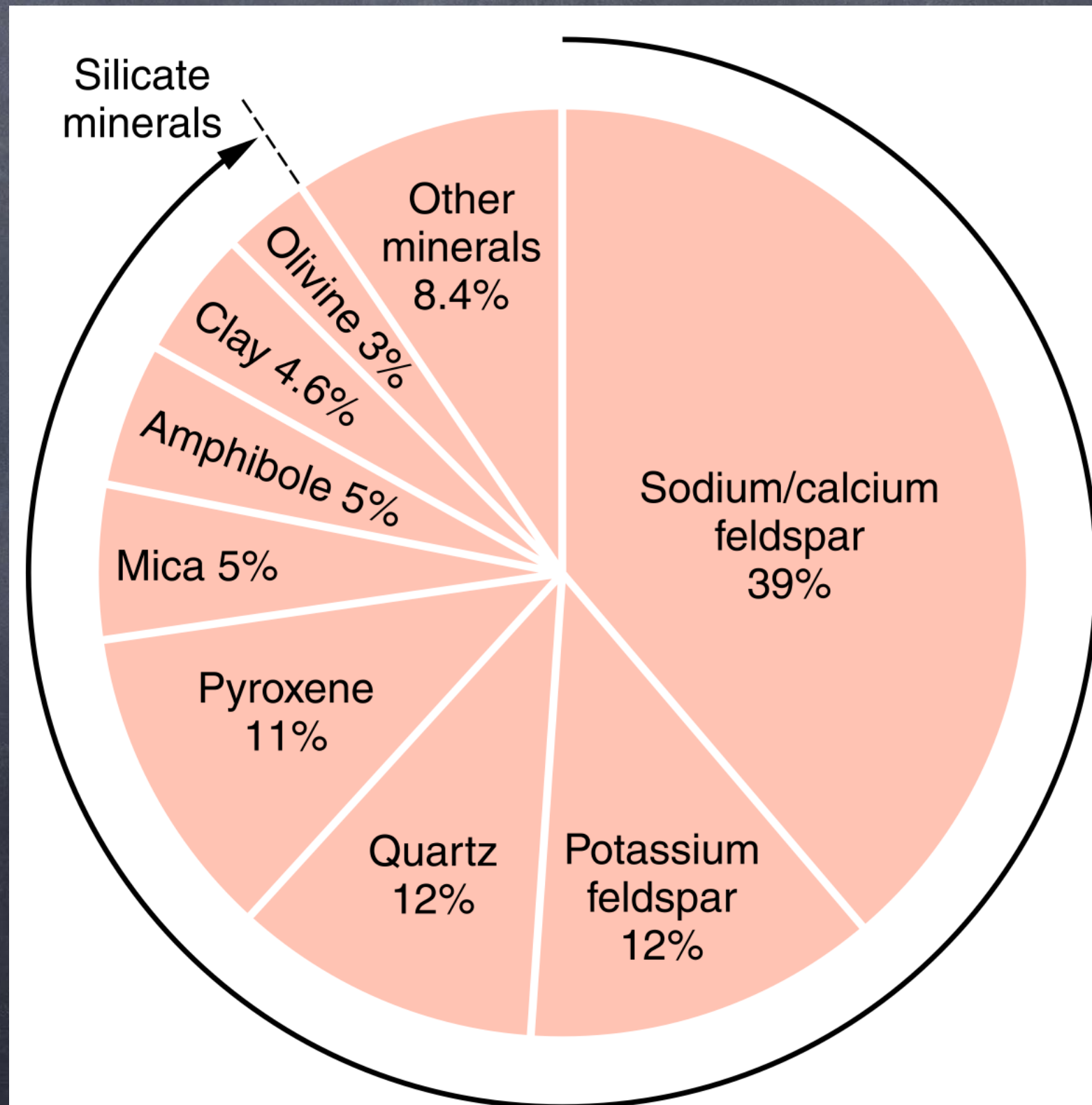
<b>ELEMENT</b> (symbol)	<b>CRUST</b>		<b>HYDROSPHERE</b>	<b>TROPOSPHERE</b>
	Percent by mass	Percent by volume	Percent by volume	Percent by volume
Oxygen (O)	46.10	94.04	33.0	21.0
Silicon (Si)	28.20	0.88		
Aluminum (Al)	8.23	0.48		
Iron (Fe)	5.63	0.49		
Calcium (Ca)	4.15	1.18		
Sodium (Na)	2.36	1.11		
Magnesium (Mg)	2.33	0.33		
Potassium (K)	2.09	1.42		
Nitrogen (N)				78.0
Hydrogen (H)			66.0	
Other	0.91	0.07	1.0	1.0



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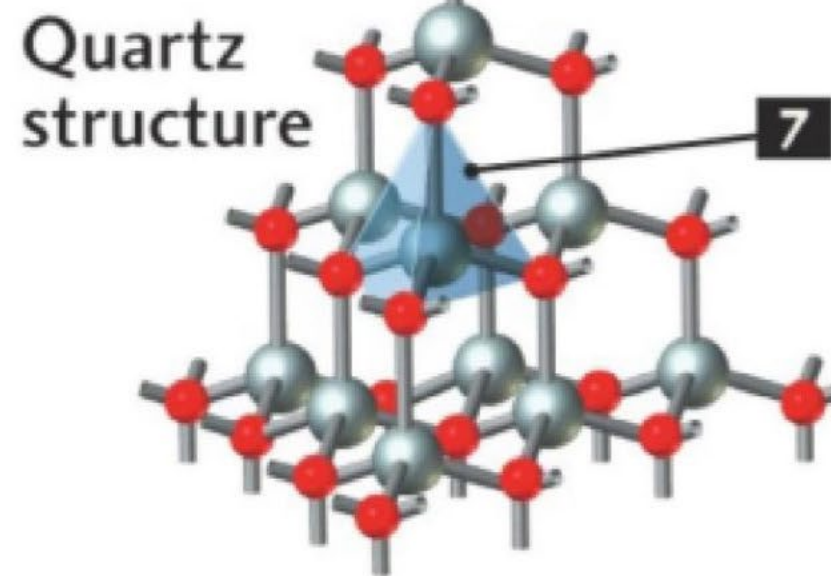
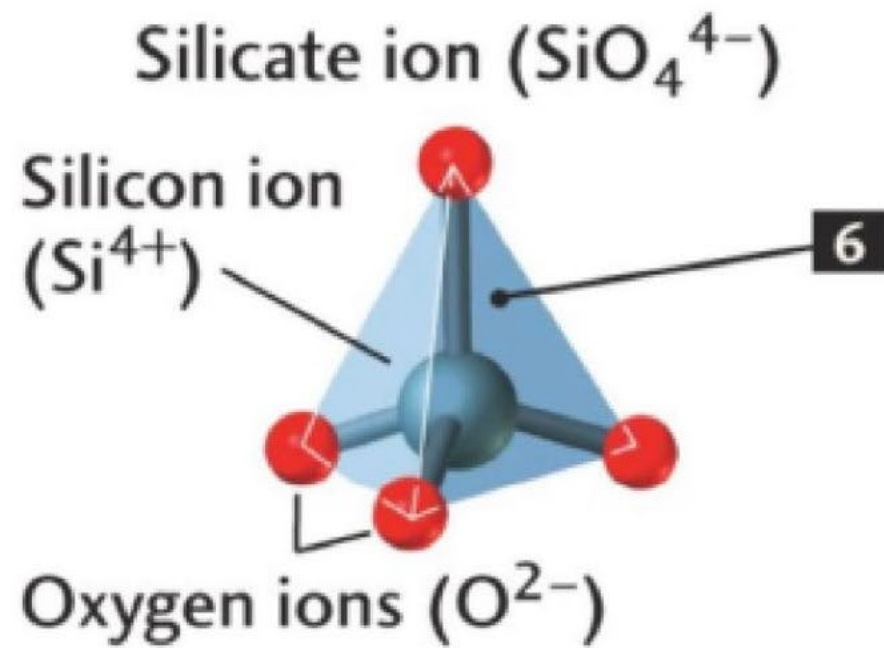
# Silicates

- The most common mineral group is the Silicates. They are made up of mostly **silicon** and **oxygen**.





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Isolated tetrahedra



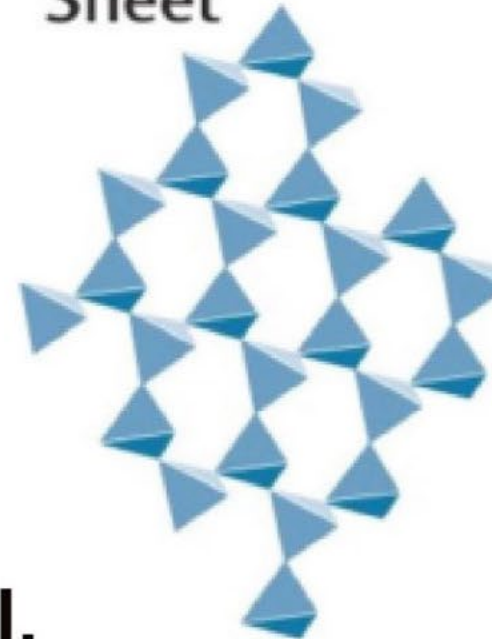
Single chains



Double chains



Sheet



Framework



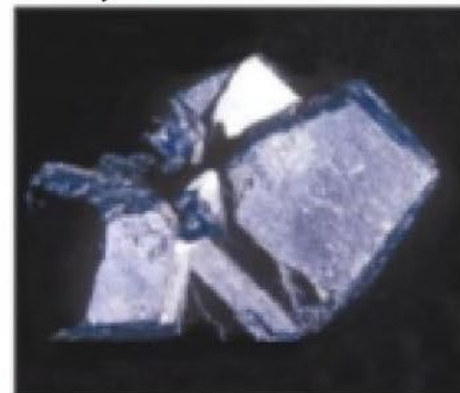
From "Understanding Earth", Press et al.



Olivine



Pyroxene



Amphibole



Muscovite



Feldspar



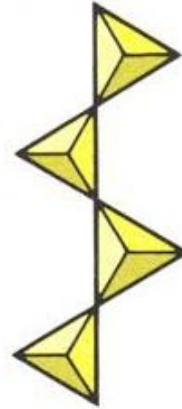
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Isolated silicate  
structure



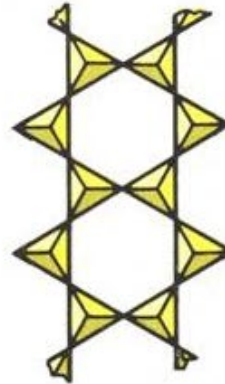
Olivine

Single chain  
structure



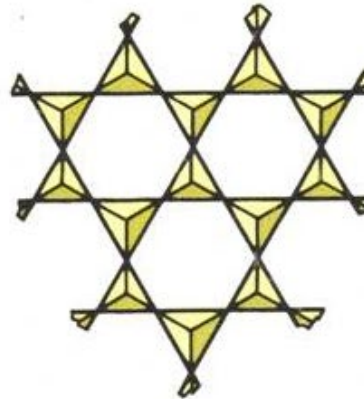
Pyroxene  
group

Double chain  
structure



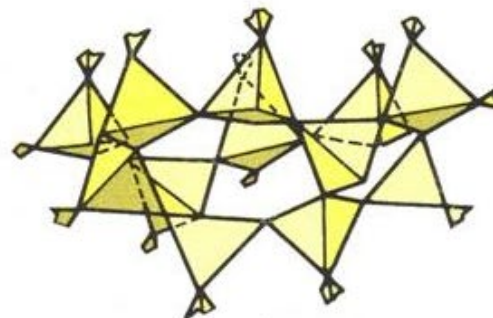
Amphibole  
group

Sheet silicate  
structure



Mica group  
Clay group

Framework silicate  
structure



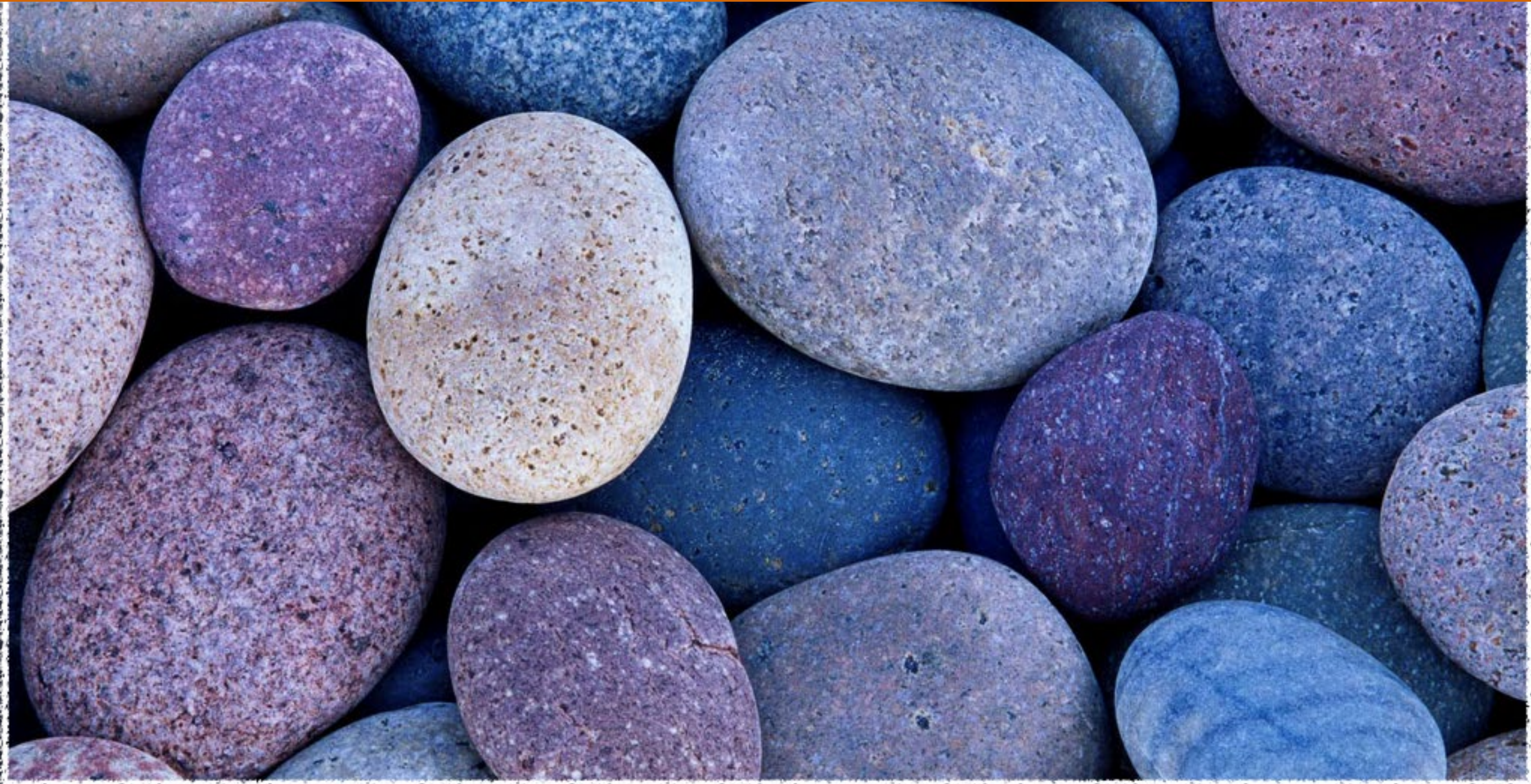
Quartz  
Feldspar group







Goal: Students will be able to identify each step of the rock cycle, and use the rock cycle chart in the ESRT.



# Rocks



Goal: Students will be able to identify each step of the rock cycle, and use the rock cycle chart in the ESRT.

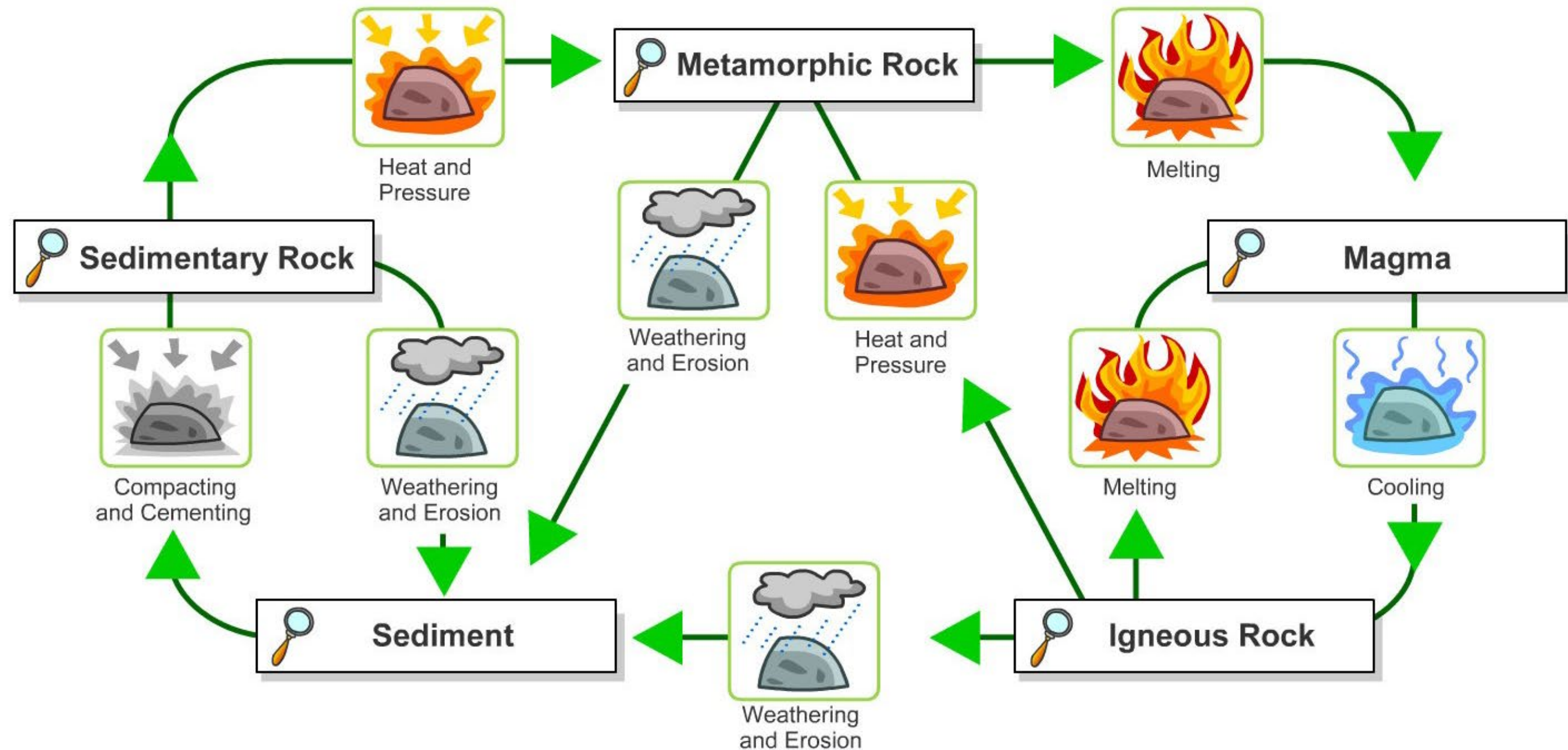
# What are rocks?

- A substance that is or was a natural part of the solid Earth, or lithosphere. Rocks are made up of a variety of minerals.
- Rocks are broken up into three groups based on how they formed.
  - Igneous
  - Sedimentary
  - Metamorphic



Goal: Students will be able to identify each step of the rock cycle, and use the rock cycle chart in the ESRT.

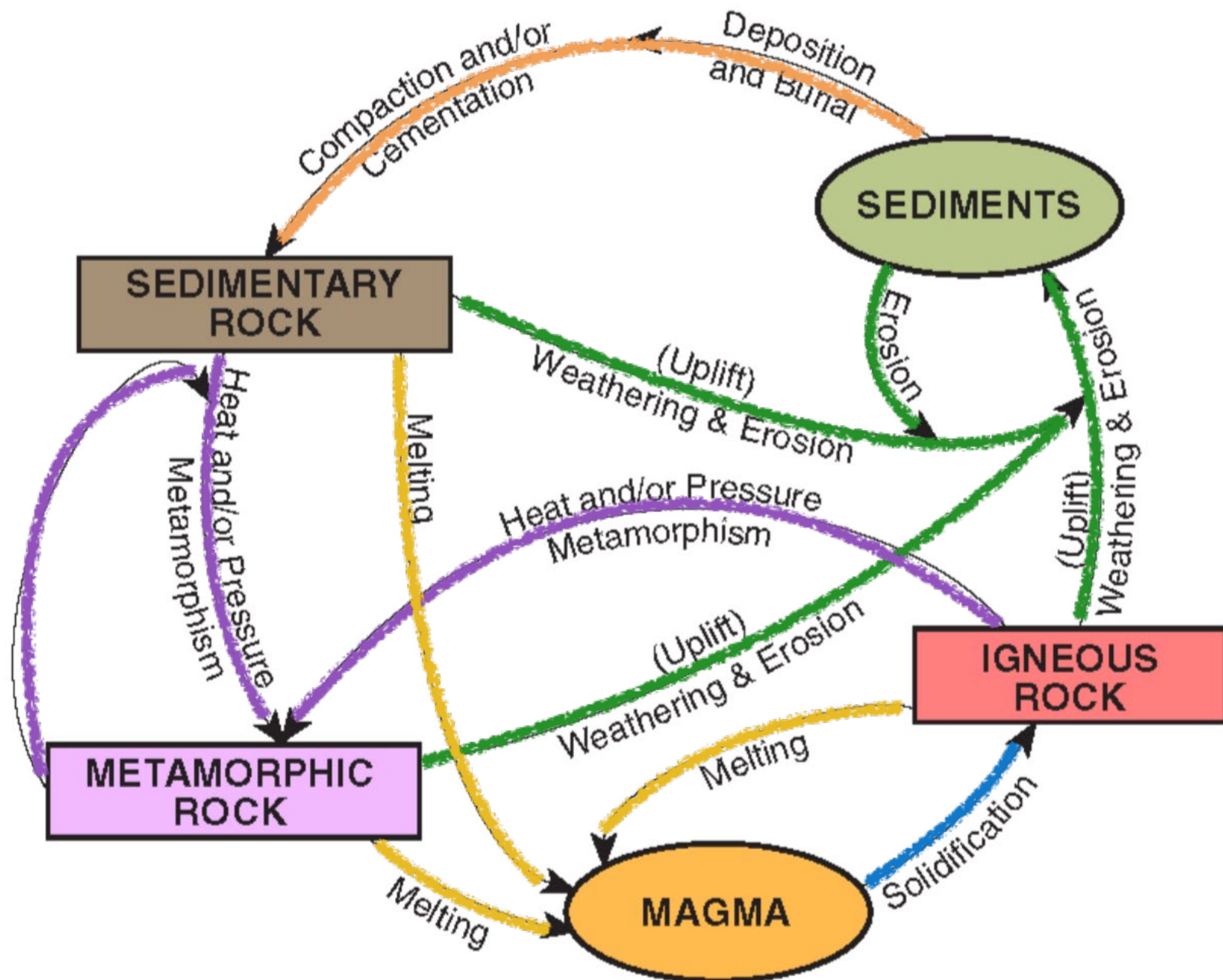
# The Rock Cycle





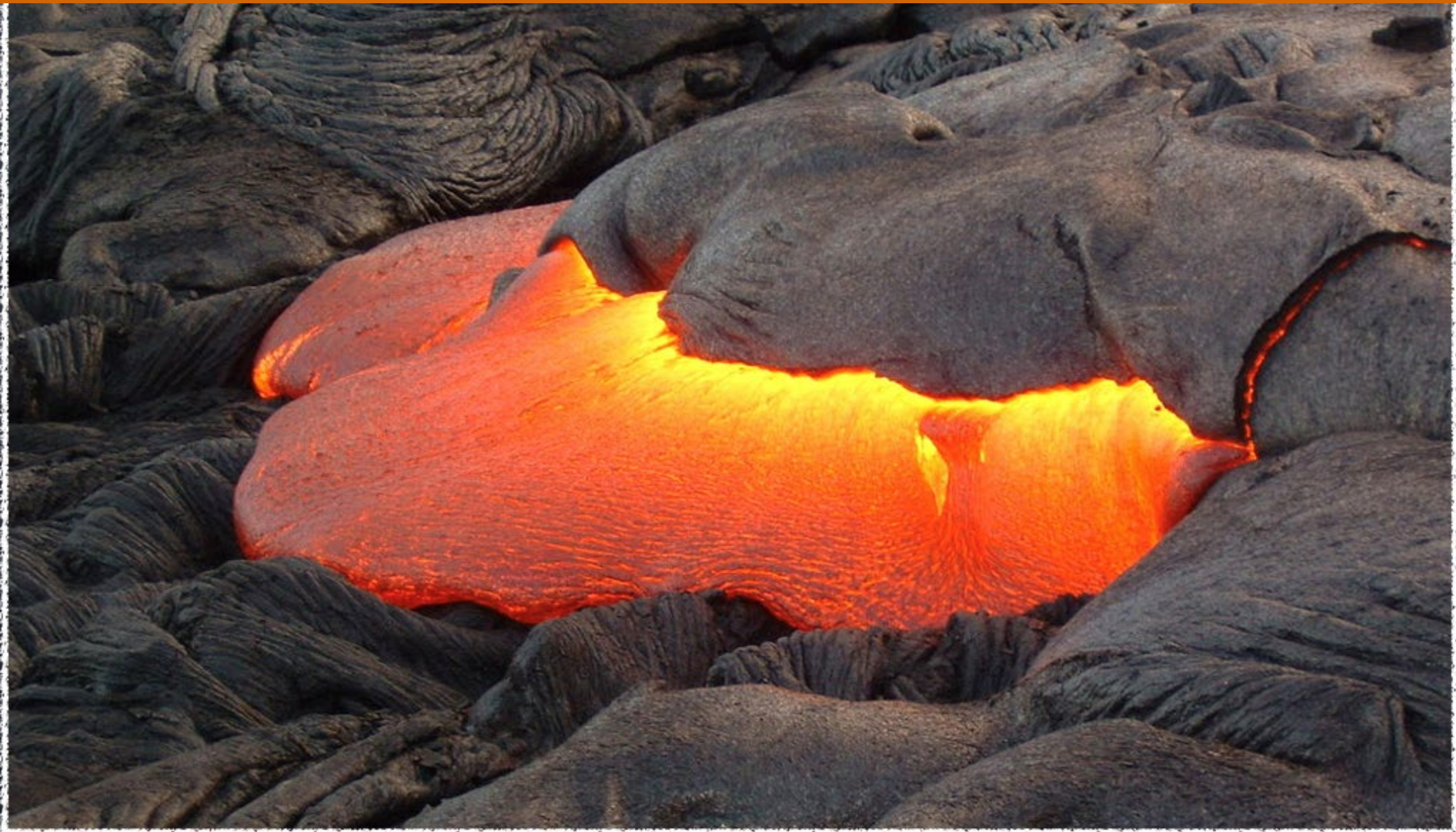
Goal: Students will be able to identify each step of the rock cycle, and use the rock cycle chart in the ESRT.

## Rock Cycle in Earth's Crust





Goal: Students will observe how rate of cooling affects mineral crystallization.



# Igneous Rocks



# Igneous Rocks

- **Igneous rocks** - form from hot, molten (liquid) rock material that originated deep within Earth.
- Liquid rock within the Earth is called **magma**.
- Liquid rock at Earth's surface is called **lava**.

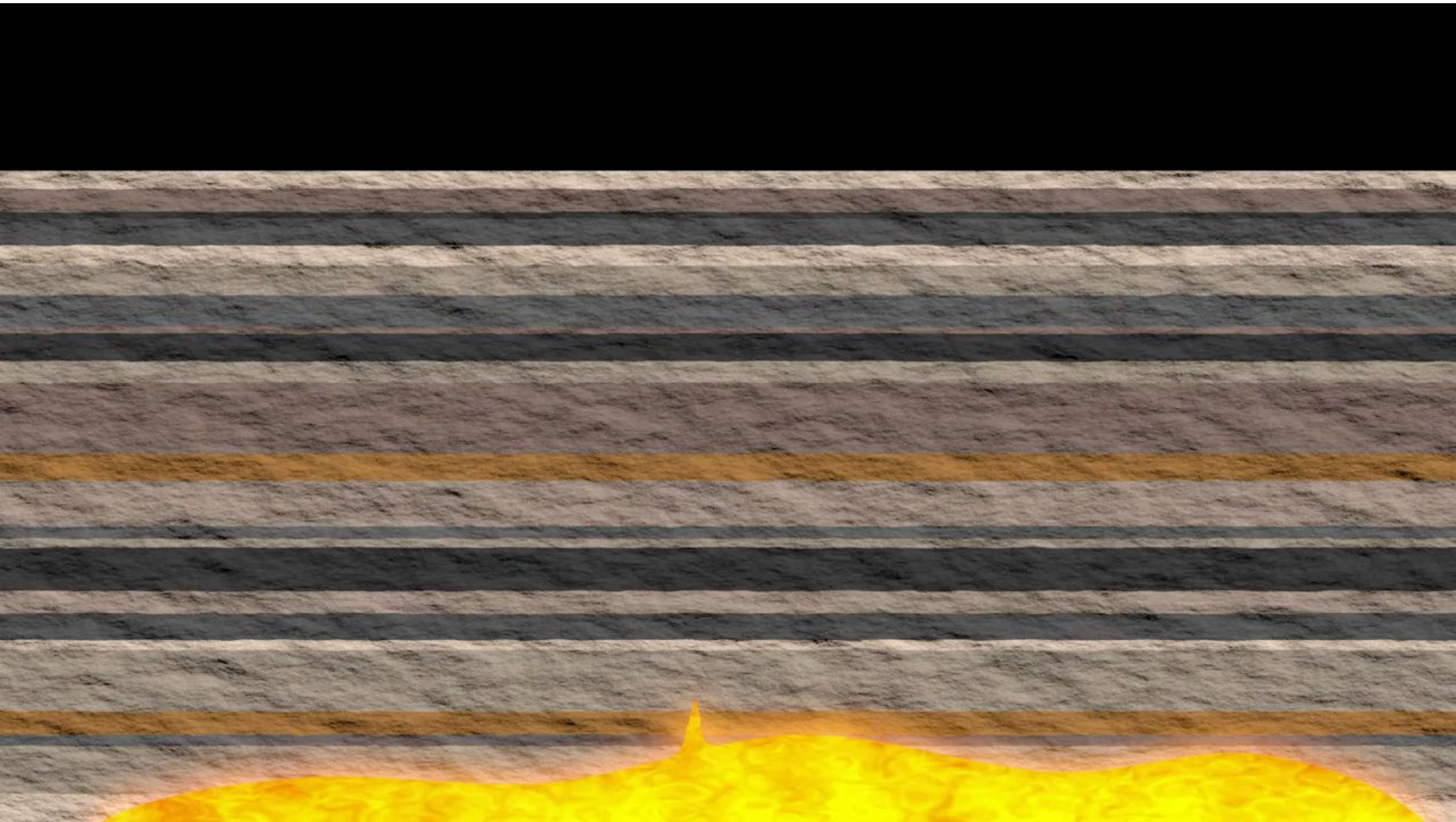


# Igneous Rocks

- Igneous rocks are broken up into two groups based on where they formed and what they are made of.
- **Intrusive** - formed inside the Earth from magma.
- **Extrusive** - formed on the Earth's crust from lava.



Creates **Intrusive** Igneous Rock since rock material never reached surface.



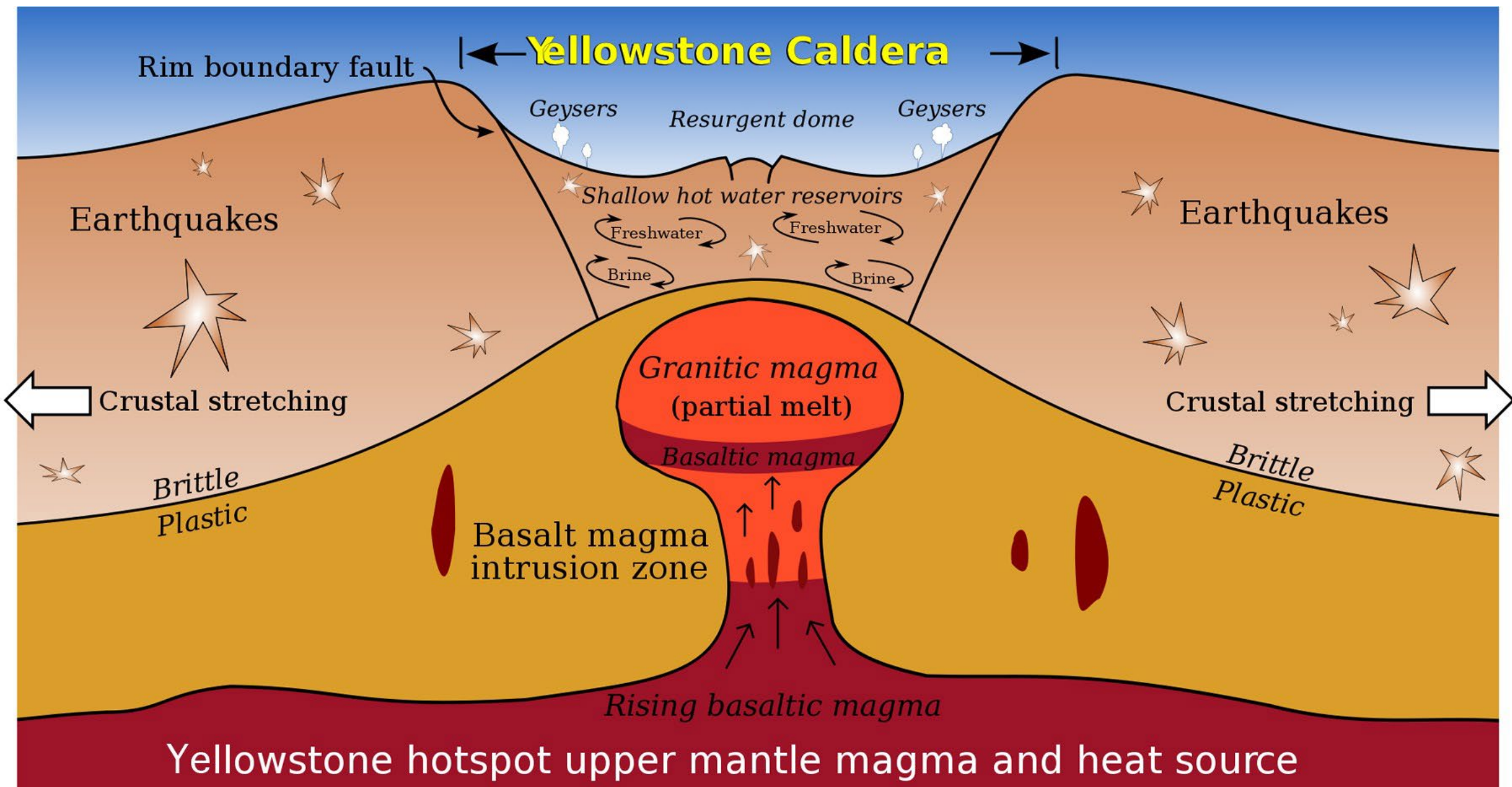


Creates **Extrusive** Igneous Rock since rock material reached surface.





Goal: Students will observe how rate of cooling affects mineral crystallization.





Goal: Students will observe how rate of cooling affects mineral crystallization.















**U.S. Geological Survey (USGS)**

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Kanaga Volcano — Here's a photo from Kanaga Volcano, one of the most southerly members of the Aleutian Islands chain. In the foreground is a stunning view of a lava flow from a 1906 event. According to records, a trapper living on the island in 1906 experienced several earthquakes and witnessed lava pouring down both east and west sides of the cone.

You can learn more about Kanaga Volcano, see images, and read about current past activity at <http://on.doi.gov/Kanaga>. Photo credit: Michelle Combs, USGS.

You can also follow USGS Volcanoes on Facebook (<https://fb.com/USGSVolcanoes/>) to learn about volcanic events, eruption anniversaries, and more.

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[#landscape](#) [#lava](#)



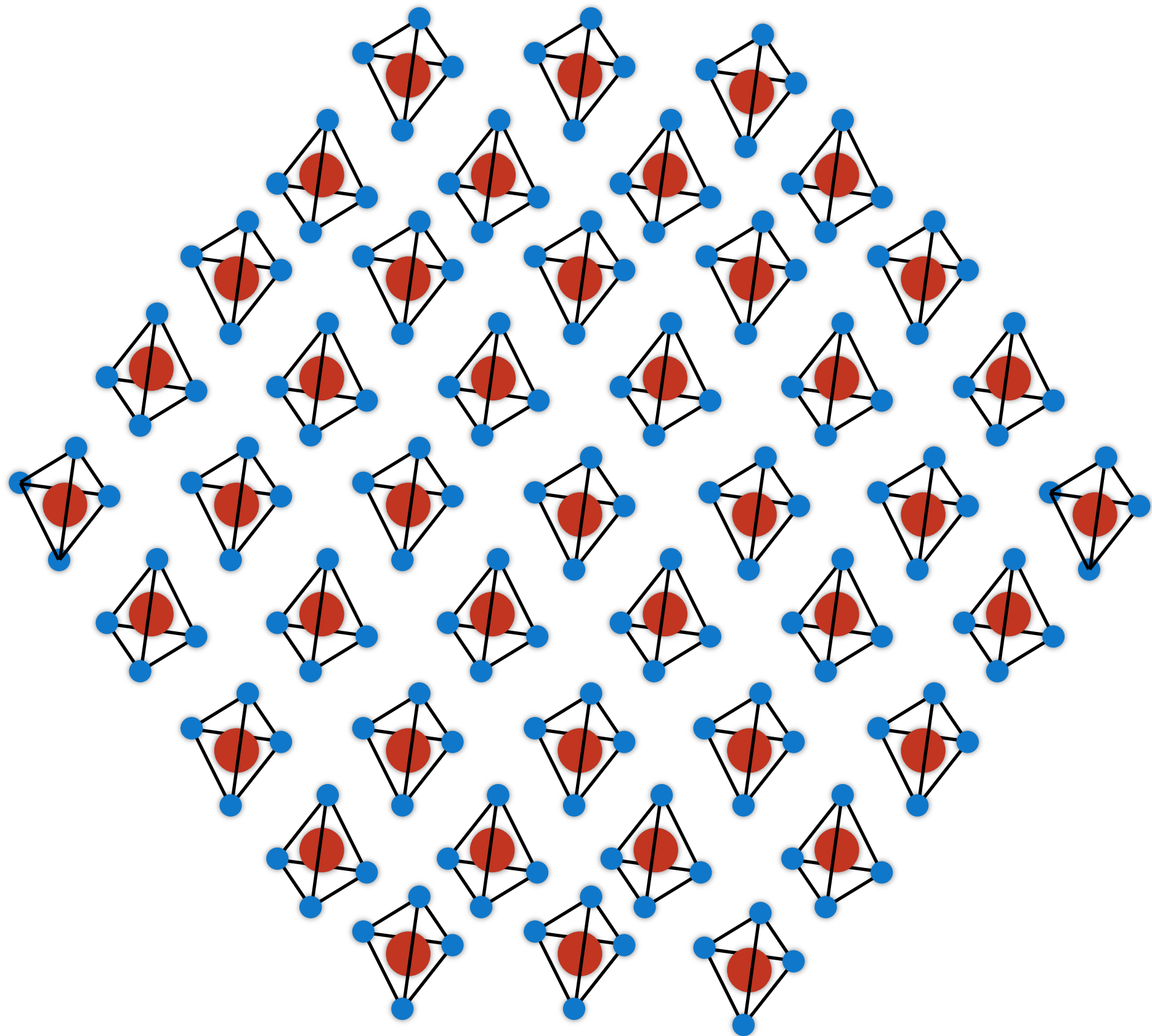
# Igneous Rocks

- Igneous rocks are broken up into two groups based on where they formed and what they are made of.
- **Intrusive** - formed inside the Earth from magma. Took a **long time to cool**, and therefore have **large crystals**.
- **Extrusive** - formed on the Earth's crust from lava. **Cooled quickly**, and therefore have **small crystals**.









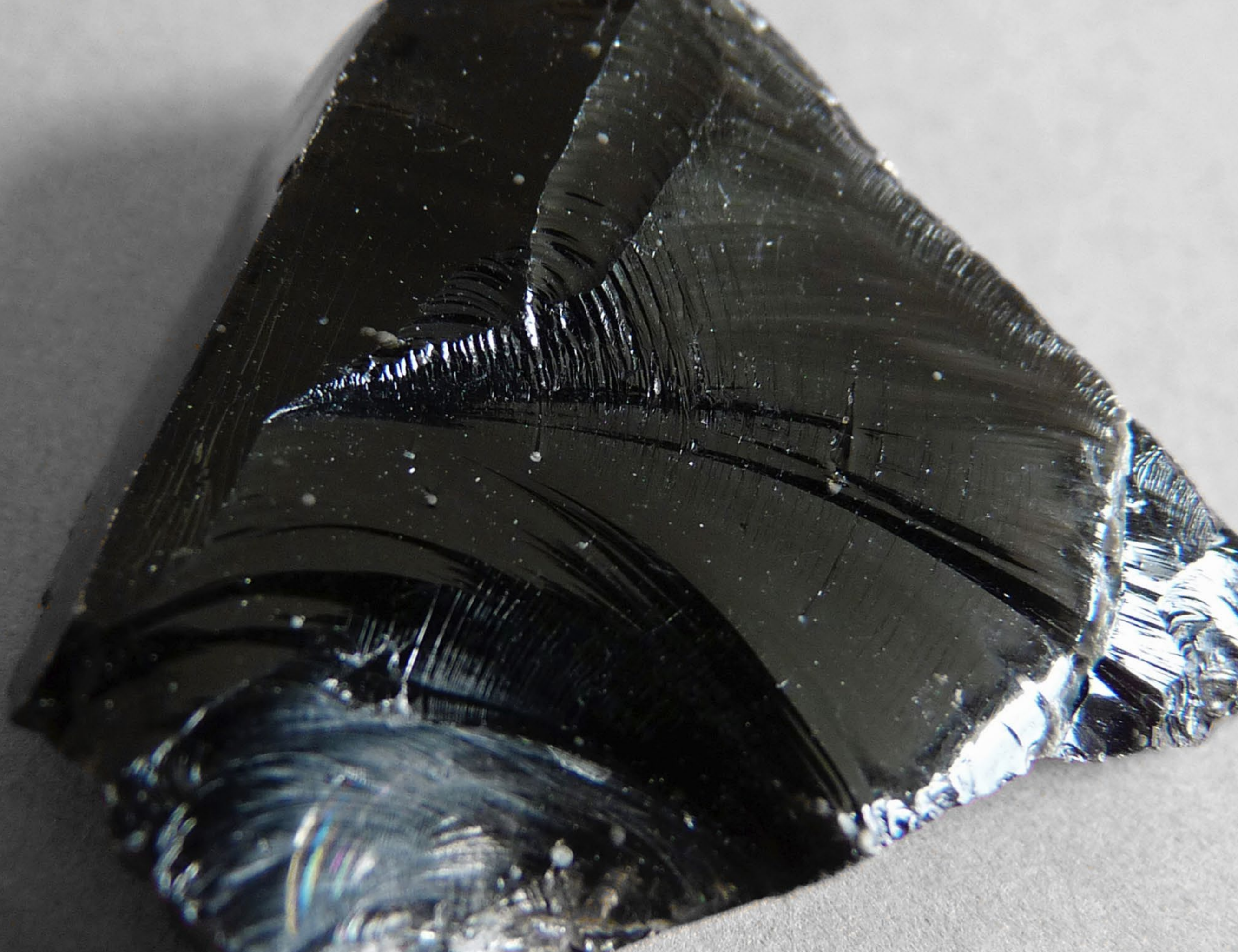
1 Million Years... 2 Million Years... 3 Million Years... 4 Million Years... 5 Million Years...



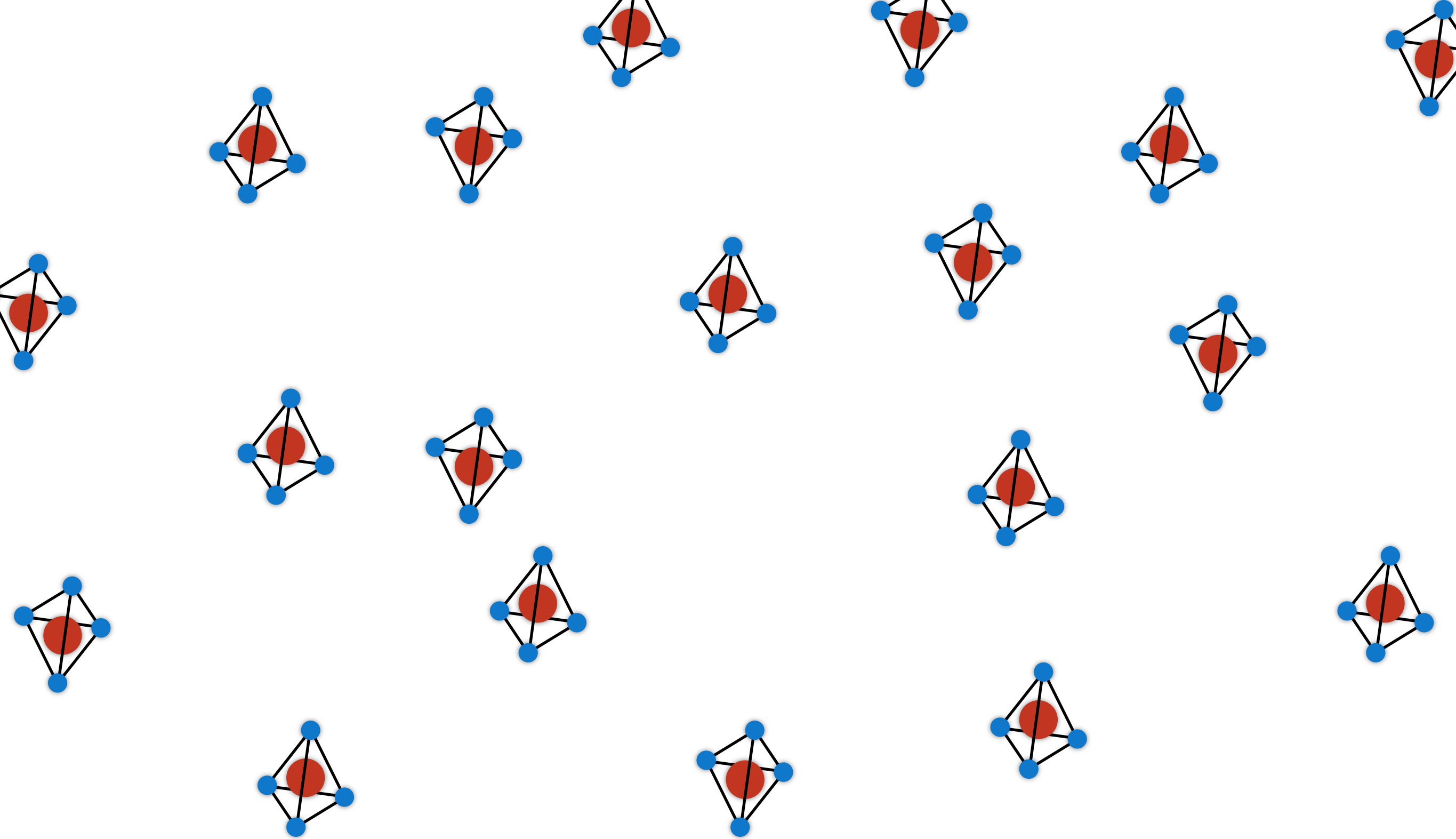


Rock took a long time to cool,  
therefore formed big  
crystals.









*A couple minutes...*





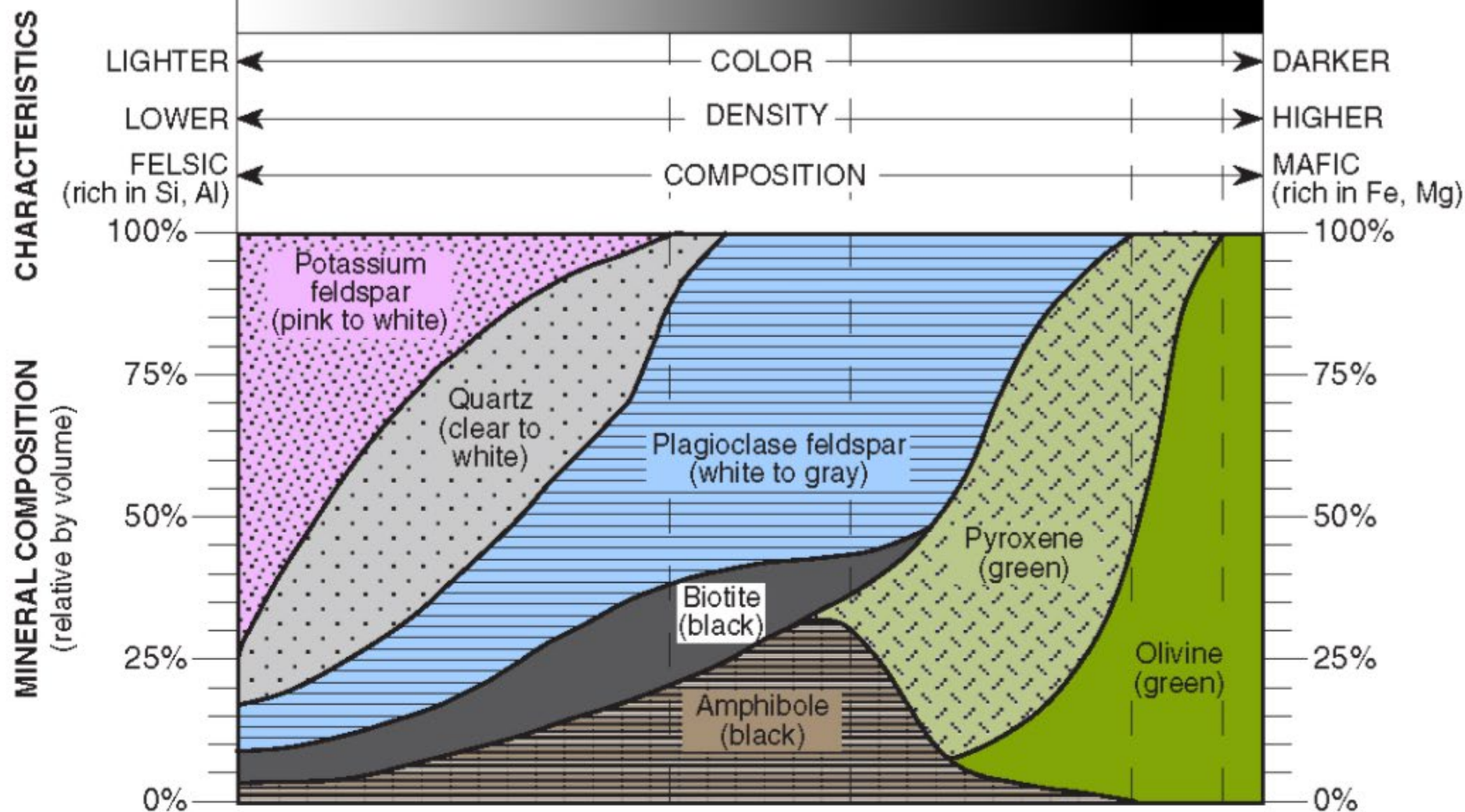
Rock cooled very rapidly,  
therefore did not form  
crystals.



Goal: Students will be able to use the "Igneous Rock Chart" in the ESRT.

## Scheme for Igneous Rock Identification

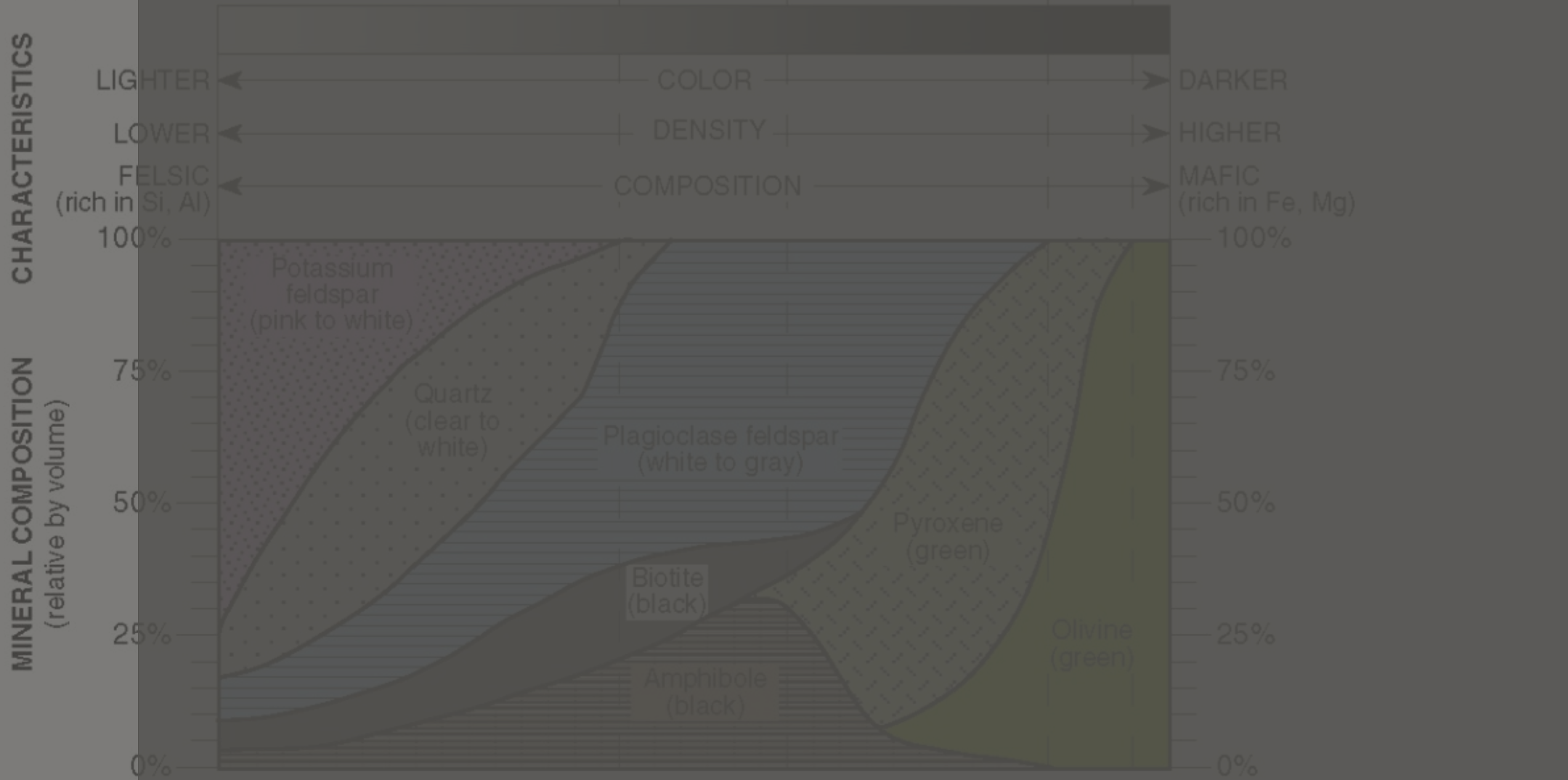
Scheme for Igneous Rock Identification							CRYSTAL SIZE	TEXTURE		
IGNEOUS ROCKS	ENVIRONMENT OF FORMATION	EXTRUSIVE (Volcanic)	Obsidian (usually appears black)		Basaltic glass		non-crystalline	Glassy	Non-vesicular	
			Pumice		Scoria				Vesicular (gas pockets)	
			Vesicular rhyolite	Vesicular andesite	Vesicular basalt		less than 1 mm	Fine		
			Rhyolite	Andesite	Basalt					
	INTRUSIVE (Plutonic)	Granite	Diorite	Diabase		Peridotite	Dunite	1 mm to 10 mm	Coarse	Non-vesicular
				Gabbro						
				Pegmatite					10 mm or larger	Very coarse





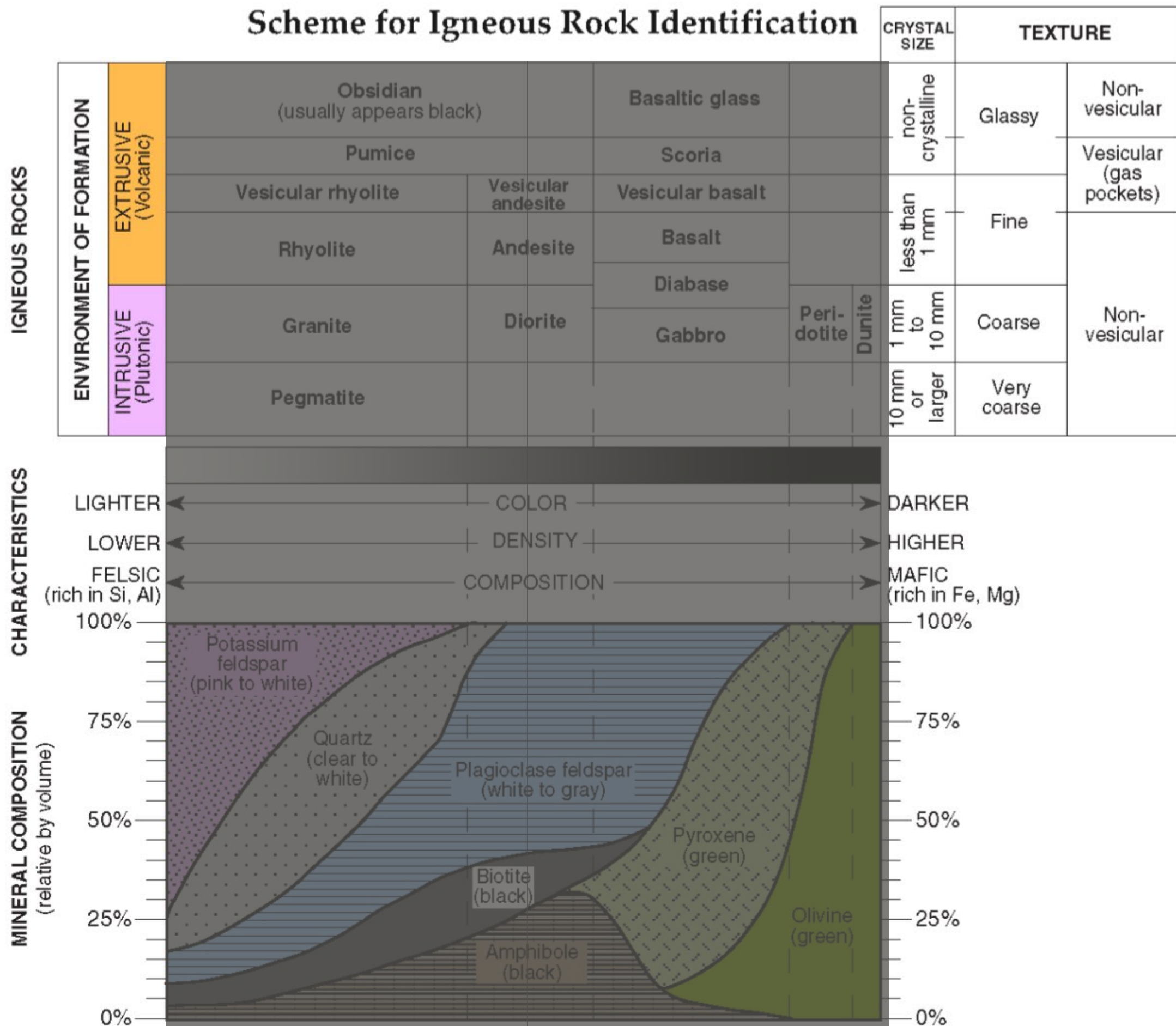
Goal: Students will be able to use the “Igneous Rock Chart” in the ESRT.

Scheme for Igneous Rock Identification						CRYSTAL SIZE	TEXTURE			
IGNEOUS ROCKS	ENVIRONMENT OF FORMATION	EXTRUSIVE (Volcanic)	Obsidian (usually appears black)		Basaltic glass		non-crystalline	Glassy	Non-vesicular	
			Pumice		Scoria				Vesicular (gas pockets)	
			Vesicular rhyolite	Vesicular andesite	Vesicular basalt		less than 1 mm	Fine	Non-vesicular	
			Rhyolite	Andesite	Basalt					
	INTRUSIVE (Plutonic)	Granite	Diorite	Diabase		Peridotite	Dunite	1 mm to 10 mm		Coarse
				Gabbro						
		Pegmatite						10 mm or larger	Very coarse	





Goal: Students will be able to use the "Igneous Rock Chart" in the ESRT.

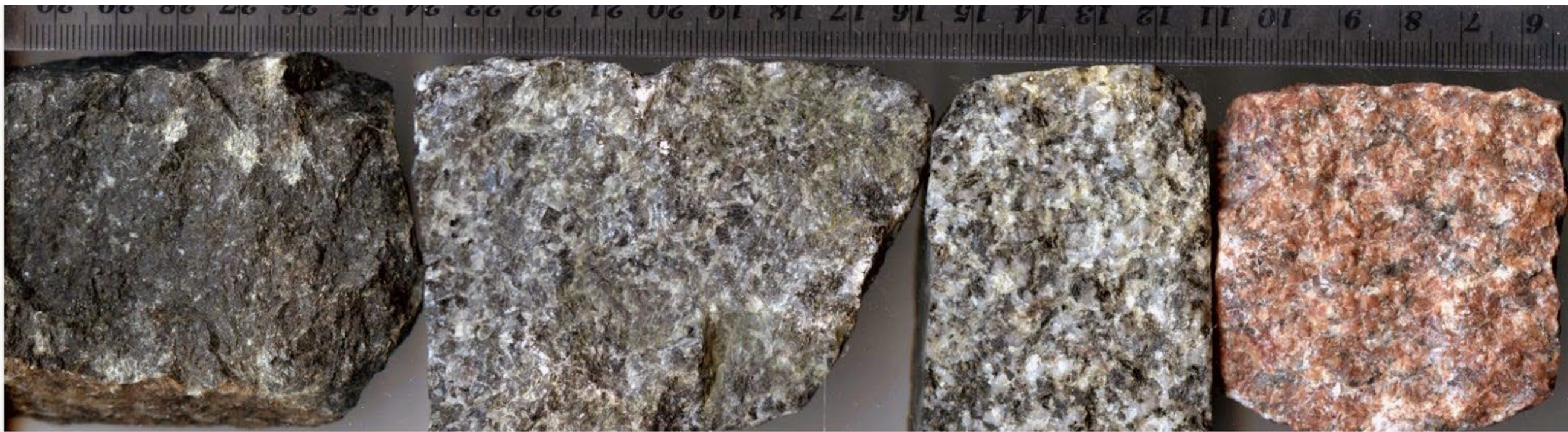




# Felsic vs. Mafic

- Felsic rocks are made of less dense material usually found at Earth's surface. They are usually lighter in color.
- Mafic rocks are made of dense material usually found within Earth. They are usually darker in color.





Ultramafic

Mafic

Intermediate

Felsic

Color

Dark



Light

SiO<sub>2</sub>

40%

50%

60%

70%

MgO

48%

15%

2.5%

0.95%

Major mineral  
content

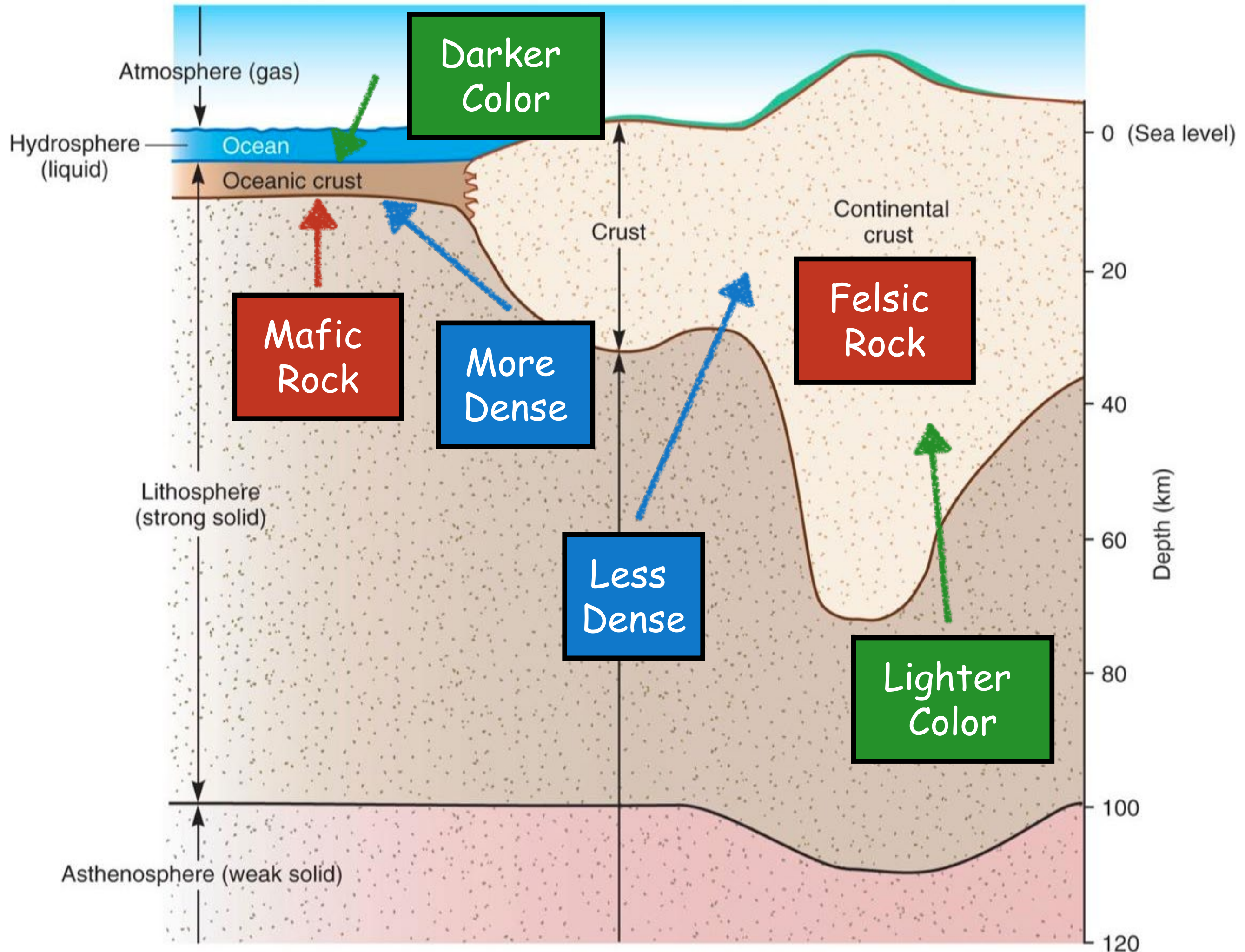
Pyroxene  
and/or olivine

Pyroxene and  
Plagioclase feldspar

Biotite  
Alkali Feldspar  
Quartz

Quartz  
Alkali Feldspar



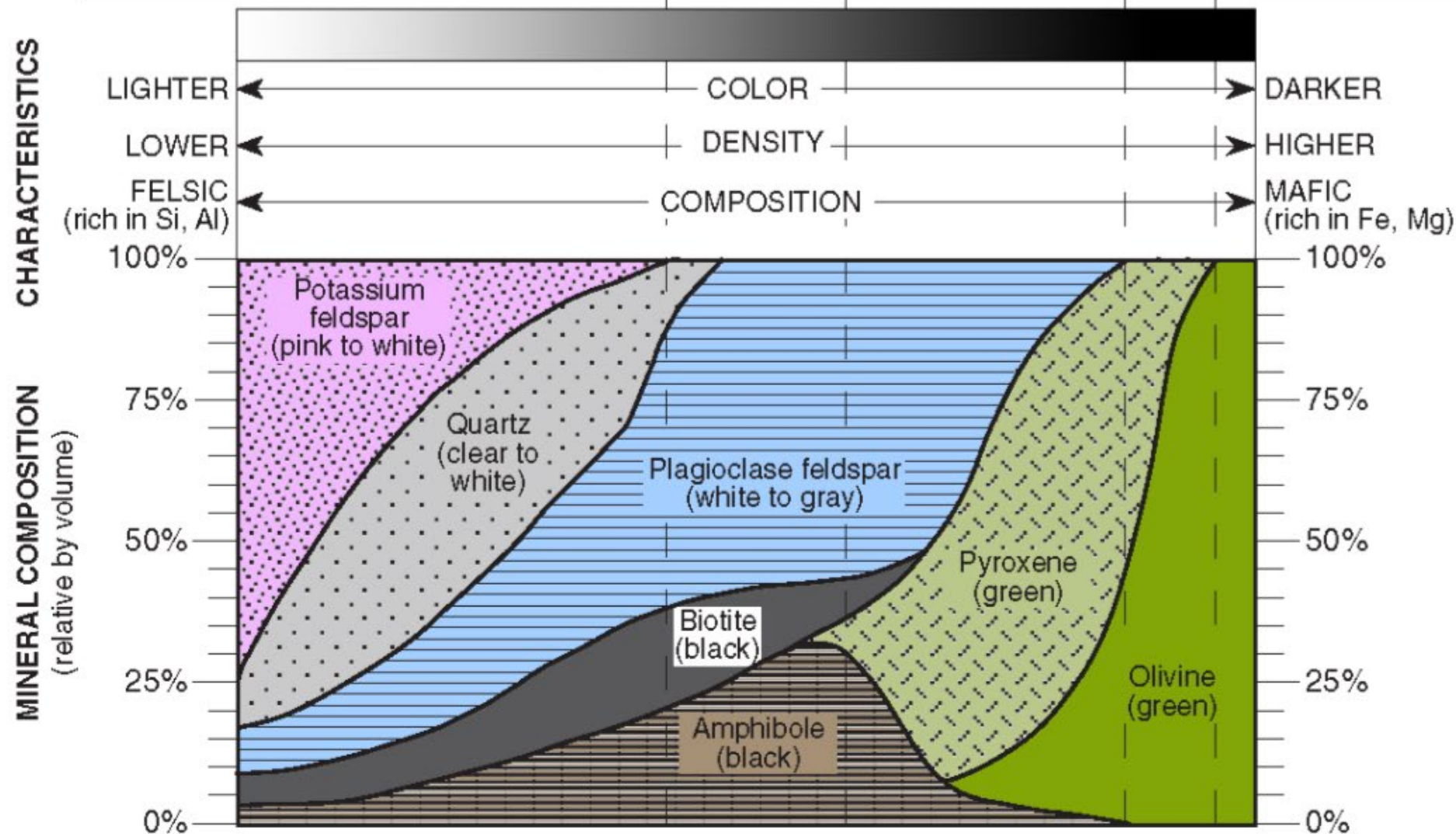




Goal: Students will be able to use the "Igneous Rock Chart" in the ESRT.

## Scheme for Igneous Rock Identification

Scheme for Igneous Rock Identification							CRYSTAL SIZE	TEXTURE	
IGNEOUS ROCKS	ENVIRONMENT OF FORMATION	EXTRUSIVE (Volcanic)	Obsidian (usually appears black)		Basaltic glass		non-crystalline	Glassy	Non-vesicular
			Pumice		Scoria				Vesicular (gas pockets)
			Vesicular rhyolite	Vesicular andesite	Vesicular basalt		less than 1 mm	Fine	Non-vesicular
			Rhyolite	Andesite	Basalt				
	INTRUSIVE (Plutonic)	Granite	Diorite	Diabase	Peridotite	Dunite	1 mm to 10 mm	Coarse	
				Gabbro					
			Pegmatite						





Goal: Students will be able to use the  
"Sedimentary Rock Chart" in the ESRT.



# Sedimentary Rocks



Goal: Students will be able to use the "Sedimentary Rock Chart" in the ESRT.

# Sedimentary Rocks

- Most sedimentary rocks are made of the **weathered** remains of other rocks that have been **eroded** and later **deposited** as **sediment** in layers.
- Others may form **chemically** or **biologically**.





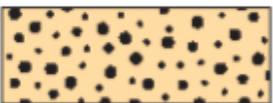


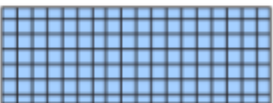
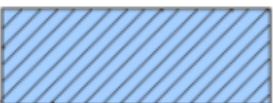

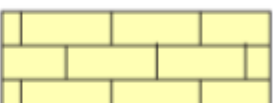

Goal: Students will be able to use the  
"Sedimentary Rock Chart" in the ESRT.





Goal: Students will be able to use the  
"Sedimentary Rock Chart" in the ESRT.

## Scheme for Sedimentary Rock Identification

INORGANIC LAND-DERIVED SEDIMENTARY ROCKS					
TEXTURE	GRAIN SIZE	COMPOSITION	COMMENTS	ROCK NAME	MAP SYMBOL
Clastic (fragmental)	Pebbles, cobbles, and/or boulders embedded in sand, silt, and/or clay	Mostly quartz, feldspar, and clay minerals; may contain fragments of other rocks and minerals	Rounded fragments	<b>Conglomerate</b>	
			Angular fragments	<b>Breccia</b>	
	Sand (0.006 to 0.2 cm)		Fine to coarse	<b>Sandstone</b>	
	Silt (0.0004 to 0.006 cm)		Very fine grain	<b>Siltstone</b>	
	Clay (less than 0.0004 cm)		Compact; may split easily	<b>Shale</b>	
CHEMICALLY AND/OR ORGANICALLY FORMED SEDIMENTARY ROCKS					
TEXTURE	GRAIN SIZE	COMPOSITION	COMMENTS	ROCK NAME	MAP SYMBOL
Crystalline	Fine to coarse crystals	Halite	Crystals from chemical precipitates and evaporites	<b>Rock salt</b>	
		Gypsum		<b>Rock gypsum</b>	
		Dolomite		<b>Dolostone</b>	
Crystalline or bioclastic	Microscopic to very coarse	Calcite	Precipitates of biologic origin or cemented shell fragments	<b>Limestone</b>	
Bioclastic		Carbon	Compacted plant remains	<b>Bituminous coal</b>	



Goal: Students will be able to use the  
"Metamorphic Rock Chart" in the ESRT.



# Metamorphic Rocks



Goal: Students will be able to use the  
"Metamorphic Rock Chart" in the ESRT.





Goal: Students will be able to use the  
"Metamorphic Rock Chart" in the ESRT.





Goal: Students will be able to use the "Metamorphic Rock Chart" in the ESRT.

# Metamorphic Rocks

- Rocks that are subjected to conditions of **heat and pressure**, but **do not melt**.
- These rocks are changed from their original form, including changes in **texture**, **mineral content**, **hardness**, **density**, as well as other visible signs of change.



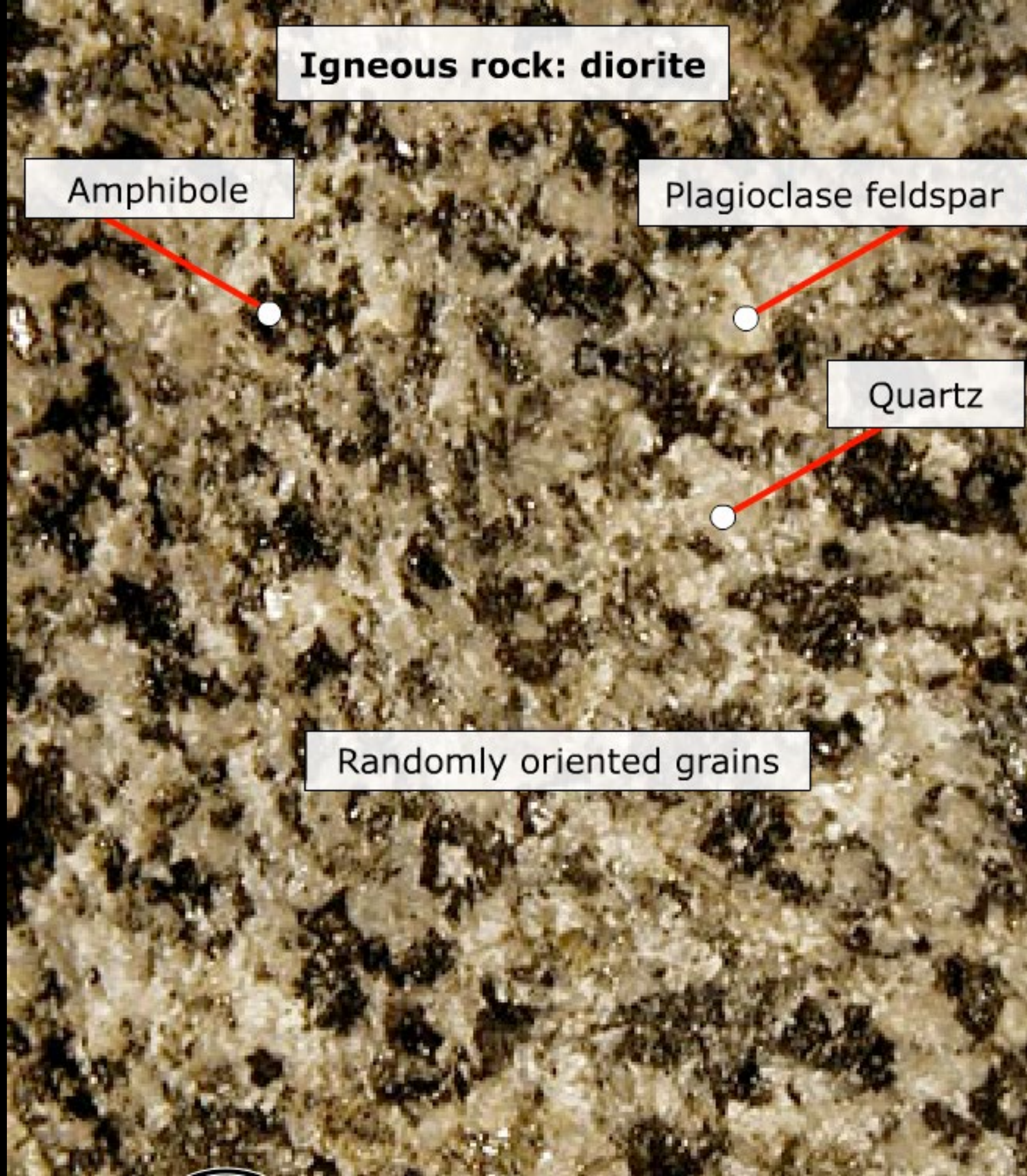
**Igneous rock: diorite**

Amphibole

Plagioclase feldspar

Quartz

Randomly oriented grains



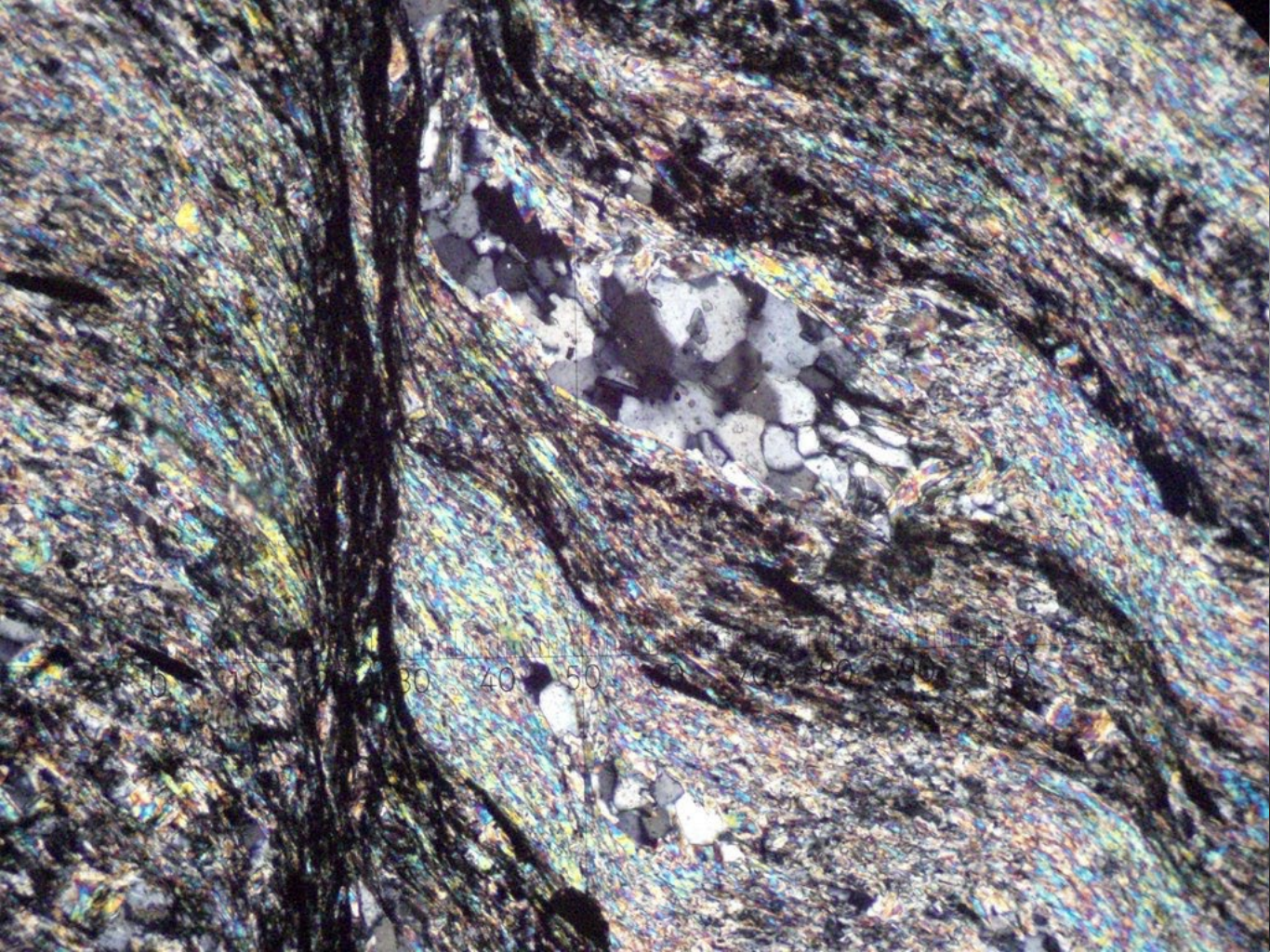


Goal: Students will be able to use the "Metamorphic Rock Chart" in the ESRT.

# Metamorphic Rocks

- Visible signs of change:
  - **Foliation** - repetitive layers of parallel crystal growth formed by intense pressure.
  - **Banding** - separation of light and dark minerals into layers.







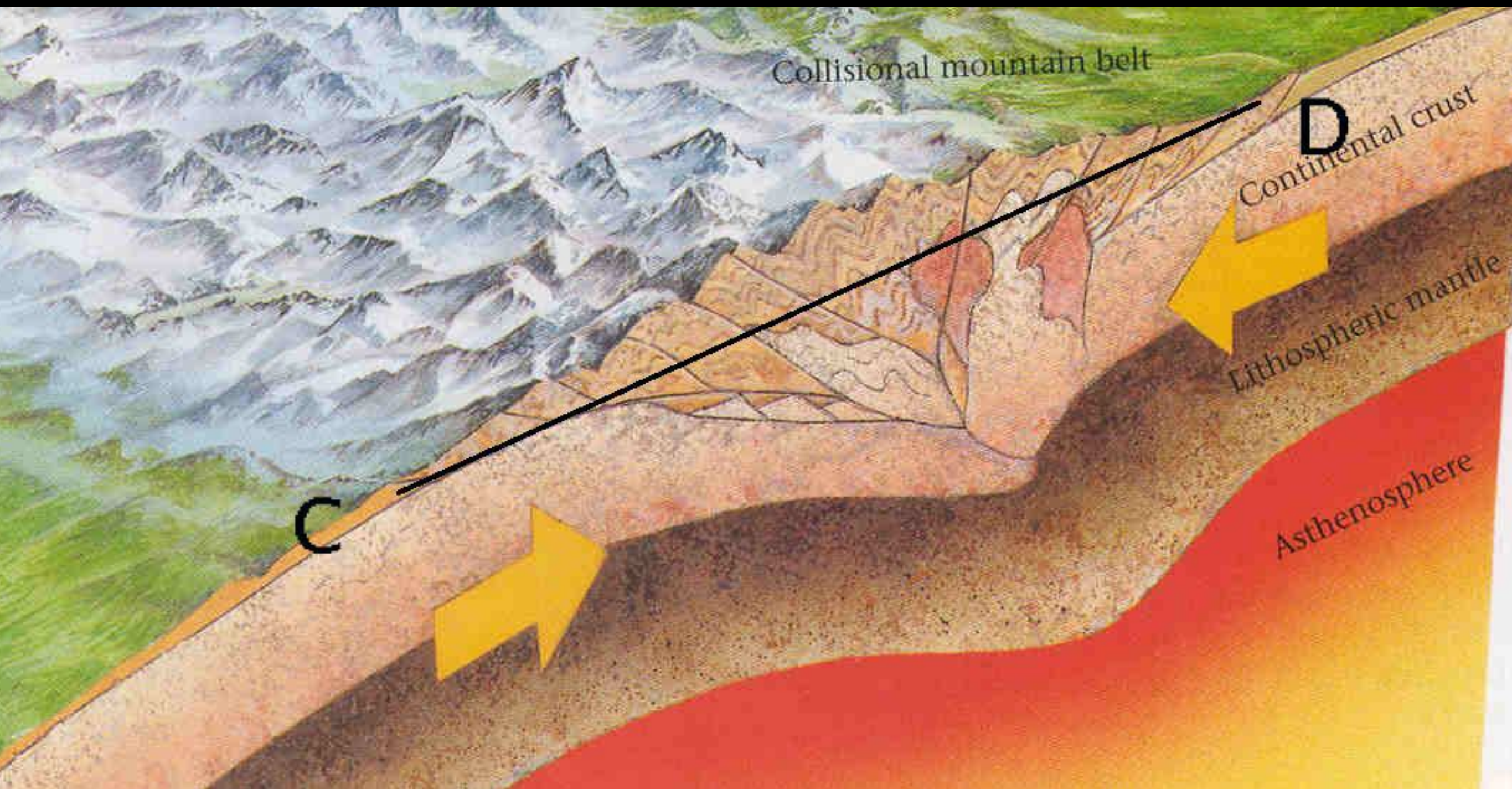
Goal: Students will be able to use the "Metamorphic Rock Chart" in the ESRT.

# How do metamorphic rocks form?

- Need intense **heat** or **pressure** to change original rock.
  - **Regional Metamorphism** - when a large mass of rock experiences intense heat and pressure.
  - **Contact Metamorphism** - rocks are exposed to intense heat, but not pressure. This happens when magma or lava touches or comes near another rock.
- **Recrystallization** - forming of different mineral crystals **without melting**. Leads to:
  - Larger crystals
  - New minerals
  - Foliation development

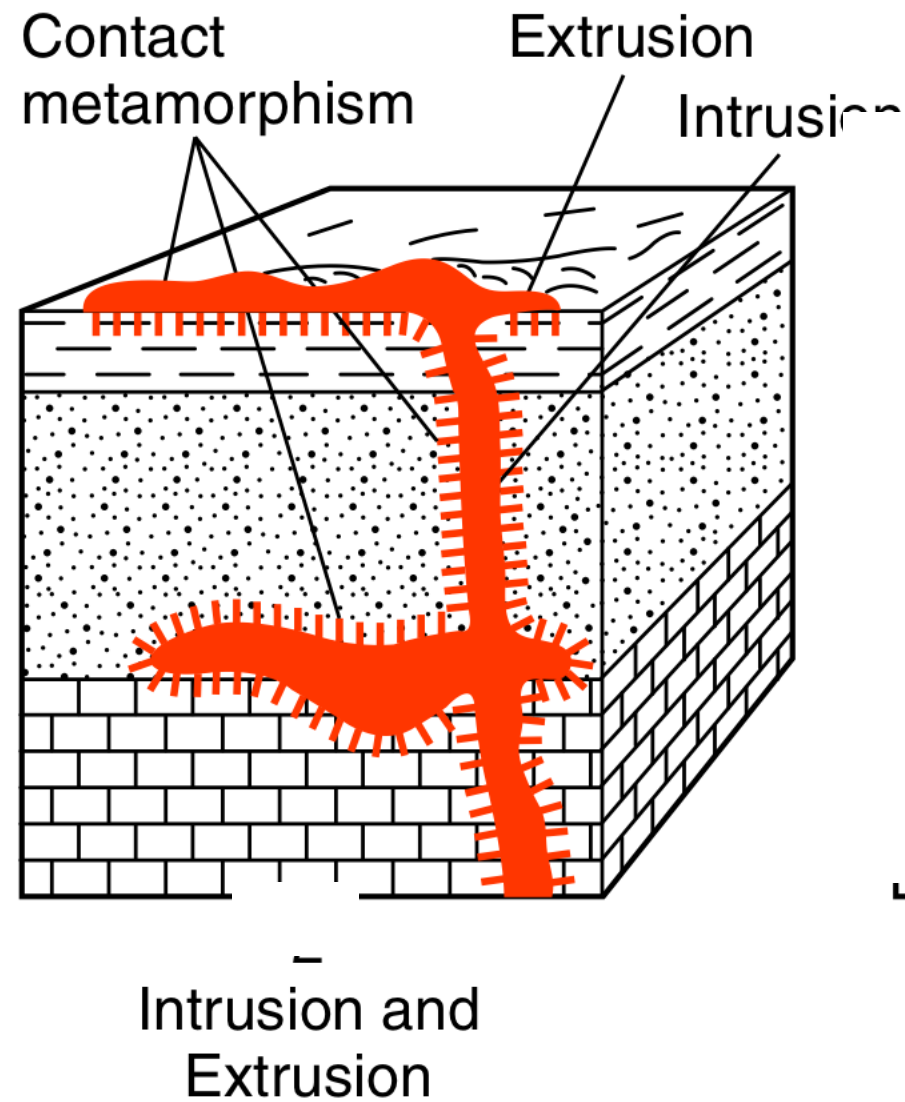


Goal: Students will be able to use the  
"Metamorphic Rock Chart" in the ESRT.

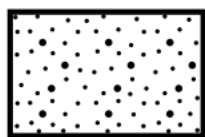




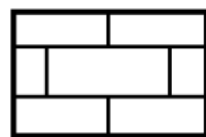
Goal: Students will be able to use the  
"Metamorphic Rock Chart" in the ESRT.



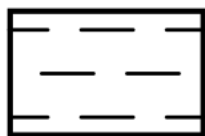
Key:



Sandstone



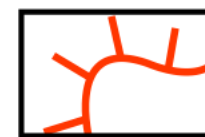
Limestone



Shale



Basalt

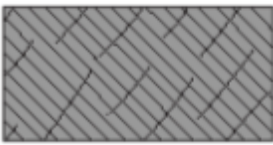


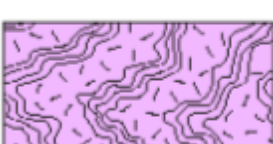

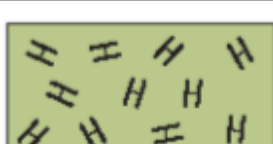
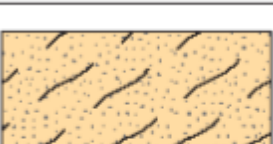

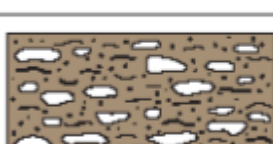


Contact  
metamorphism  
(transition zone)



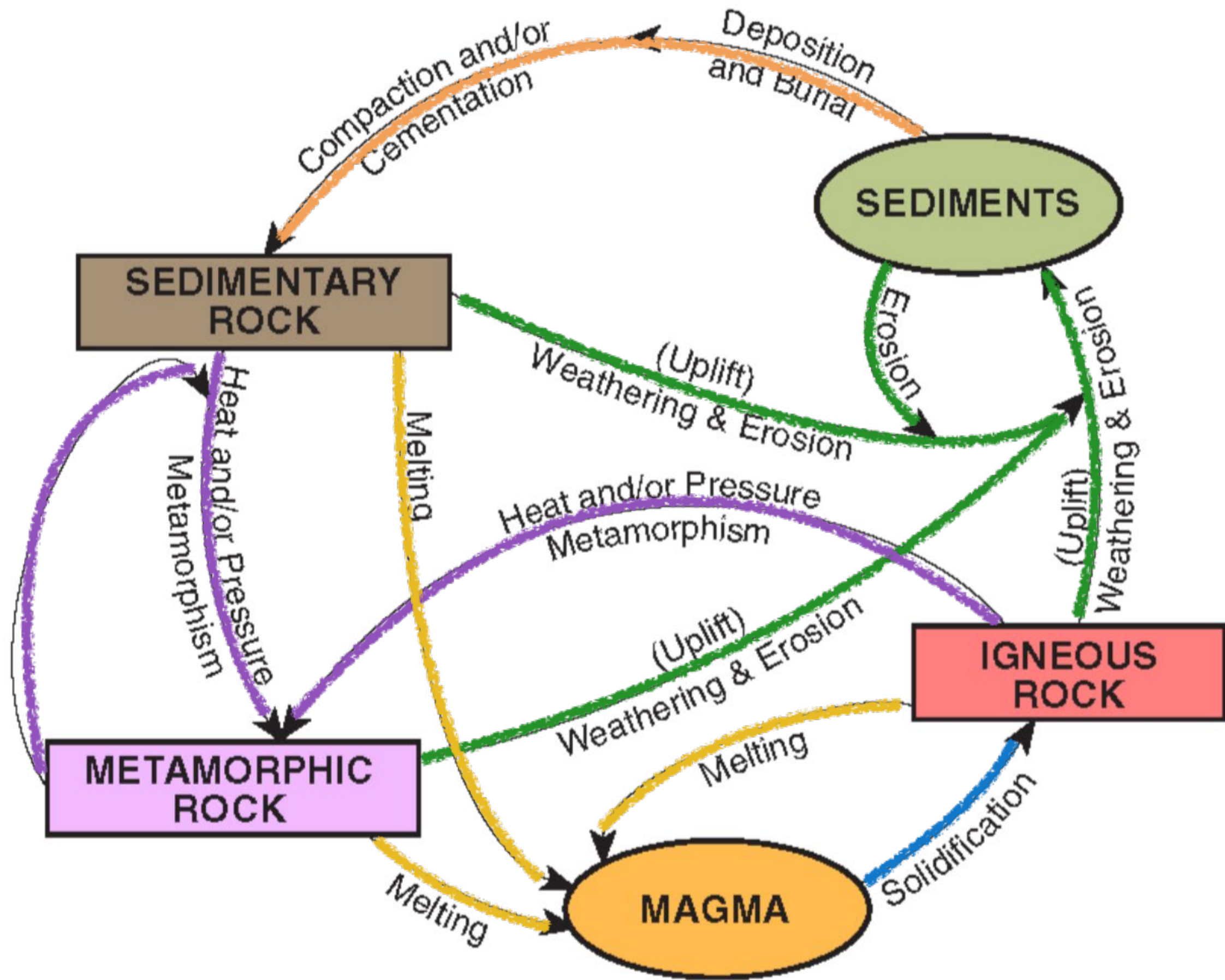
Goal: Students will be able to use the  
"Metamorphic Rock Chart" in the ESRT.

## Scheme for Metamorphic Rock Identification

TEXTURE	GRAIN SIZE	COMPOSITION	TYPE OF METAMORPHISM	COMMENTS	ROCK NAME	MAP SYMBOL
FOLIATED	MINERAL ALIGNMENT		Regional (Heat and pressure increases)	Low-grade metamorphism of shale	Slate	
				Foliation surfaces shiny from microscopic mica crystals	Phyllite	
				Platy mica crystals visible from metamorphism of clay or feldspars	Schist	
	BAND-ING			High-grade metamorphism; mineral types segregated into bands	Gneiss	
NONFOLIATED	Fine	Carbon	Regional	Metamorphism of bituminous coal	Anthracite coal	
	Fine	Various minerals	Contact (heat)	Various rocks changed by heat from nearby magma/lava	Hornfels	
	Fine to coarse	Quartz	Regional or contact	Metamorphism of quartz sandstone	Quartzite	
		Calcite and/or dolomite		Metamorphism of limestone or dolostone	Marble	
	Coarse	Various minerals		Pebbles may be distorted or stretched	Metaconglomerate	



# Rock Cycle in Earth's Crust





# Rocks and Minerals